

**Chemistry 12  
Homework Assignment 3-1**

46

Name KEY

Due Date \_\_\_\_\_

**Correct and Hand In By** \_\_\_\_\_

*This worksheet covers material from Tutorials 7 and 8, class notes and SW pages 73-87.*

**These questions will be marked and the mark recorded in the "Homework" section in the mark book. One set of corrections will be allowed. Get help if you need it.**

1. Decide whether each of the following are electrolytes or not (yes or no) when dissolved in water. (5 marks)

5

- a)  $\text{AgNO}_3$  ..... yes
- b)  $\text{CH}_3\text{CH}_2\text{OH}$  ..... no
- c)  $\text{NH}_4\text{Cl}$  ..... yes
- d)  $\text{CH}_3\text{CH}_2\text{COOH}$  ..... yes
- e)  $\text{Ba}(\text{OH})_2$  ..... yes

2. Give three examples of ionic compounds (formulas must be correct) (3 marks)

3

- 1. \_\_\_\_\_
- 2. \_\_\_\_\_
- 3. \_\_\_\_\_

3. Give two examples of molecular compounds (formulas must be correct) (2 marks)

2

- 1. \_\_\_\_\_
- 2. \_\_\_\_\_

Define a saturated solution (1 mark) \_\_\_\_\_

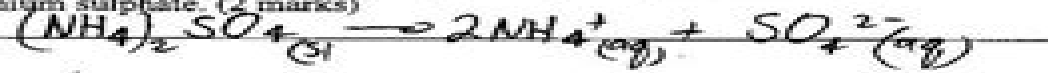
1

Solubility is the equilibrium concentration of a substance in solution at a given temperature. (1 mark)

2

6. Write a net ionic equation (balanced, all charges, subscripts, arrow(s) correct, correct order etc.) for dissolving ammonium sulphate. (2 marks)

14  
14



# Chemistry Unit 3 Test Answer Key

**Gretchen Bitterlin, Dennis  
Johnson, Donna Price, Sylvia Ramirez**

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