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# **Quantitative Methods For Business 13th Edition** **Solution Manual**

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# Quantitative Methods For Business 12th Edition Answers

**D Kirk**



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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the unknown slat burned orange but also had a slight blue color close to the stick

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spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material

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metal cations in this lab how do the metal cations become excited

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is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of

various elements in gas discharge tubes

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coefficient  $v$  shall be taken from table 7 depending upon the parameter  $e_l$  as given in 8.8

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