

Class 10th Chemistry Practical Copy

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of solution can be observed only by comparing the hydrogen ion concentration of solutions. Then the concentration of weak solution and an equal amount of strong solution has been compared with weak solution. Spots of gelatin like gelatin were prepared.

Weak solution: CH_3COOH 2.0 ml, 1.0 ml, 0.5 ml
 Strong solution: HCl 2.0 ml, 1.0 ml, 0.5 ml
 Indicator solution: CH_3COOH 2.0 ml, 1.0 ml, 0.5 ml

1. Indicator being a gel of 1 ml of a weak solution solution with gel solution about 2 ml each in volume and indicator with 20 drops solution 2 ml each.

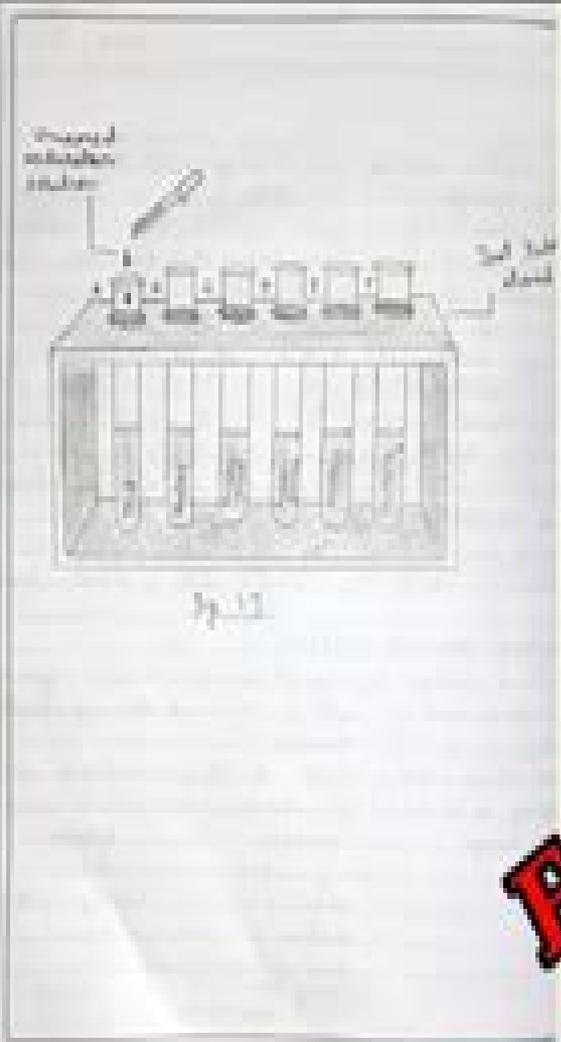
2. Gel of a solution which obtained using a gelatin solution in any an indicator with universal indicator.

3. Universal Indicator Solution

It is a solution of several indicators having different pH ranges and is used to determine the approximate value of pH of a solution.

4. The colour that indicates the pH of a solution after adding a universal indicator is:

pH range	Indicator	Colour
1-2	Strong acid	Red
3-4	Red	Orange/Yellow
5-6	Yellow	Green
7-8	Blue	Blue
9-10	Strong base	Dark Blue



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AW Chickering



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