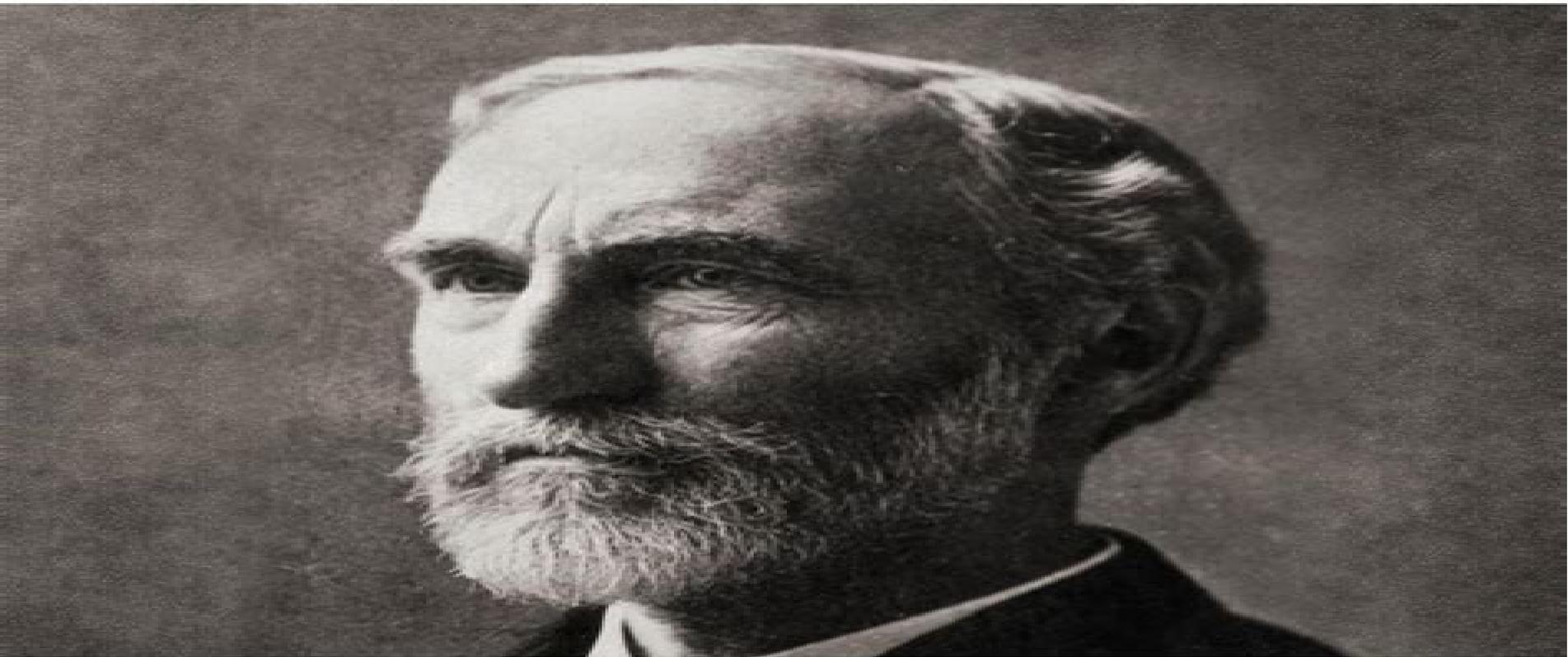


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however it is desirable to take a broader view of the subject We may imagine a great number of systems of the same nature but differing in the configurations and velocities which they have at a given instant and differing not merely infinitesimally but it may be so as to embrace every conceivable combination of configuration and velocities And here we may set the problem not to follow a particular system through its succession of configurations but to determine how the whole number of systems will be distributed among the various conceivable configurations and velocities at any required time when the distribution has been given for some one time The fundamental equation for this inquiry is that which gives the rate of change of the number of systems which fall within any infinitesimal limits of configuration and velocity About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

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chemistry texts is of necessity compressed to the point where the less competent student is unable to appreciate or comprehend its logic and beauty and is reduced to memorizing a series of formulas. It has been my aim to fill this need by writing a logical account of the foundations and applications of the subject at a level which can be grasped by an undergraduate who has had some exposure to calculus and to the basic concepts of classical thermodynamics. It can serve as a text or supplementary reading for a course or provide the means whereby one could become conversant with the subject on his own without the benefit of an instructor.

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