

## Engineering Thermodynamics

### Unit 4 and Unit 5

#### Question Bank

#### Part A Questions

1. The percentage of Oxygen present in atmospheric air on mass basis  
(a) 21                      (b) 23                      (c) 79                      (d) 77
2. The percentage of Nitrogen present in atmospheric air on mass basis  
(a) 21                      (b) 23                      (c) 79                      (d) 77
3. The percentage of Oxygen present in atmospheric air on mass basis  
(a) 21                      (b) 23                      (c) 79                      (d) 77
4. The percentage of Nitrogen present in atmospheric air on mass basis  
(a) 21                      (b) 23                      (c) 79                      (d) 77
5. The temperature at which water vapour starts condensing is called  
(a) Dry bulb temperature                      (b) Wet bulb temperature  
(c) Dew point temperature                      (d) Boiling temperature
6. Psychrometer measures  
(a) Dry bulb temperature                      (b) Wet bulb temperature  
(c) Dew point temperature                      (d) Both DBT and WBT
7. The X- axis of the psychrometric chart represents  
(a) Dry bulb temperature                      (b) Wet bulb temperature  
(c) Specific humidity                      (d) Relative humidity
8. Which one of the following remains constant during sensible cooling process?  
(a) Enthalpy (h)                      (b) Dry bulb temperature  
(c) Specific humidity (W)                      (d) Relative humidity ( $\phi$ )
9. Which one of the following remains constant during sensible heating process?  
(a) Enthalpy (h)                      (b) Dry bulb temperature  
(c) Specific humidity (W)                      (d) Relative humidity ( $\phi$ )
10. The relative humidity  $\phi$  in percentage for dry air is  
(a) 0                      (b) 25                      (c) 75                      (d) 100

# Engineering Thermodynamics Question Bank

**K Morrison**



## **Engineering Thermodynamics Question Bank:**

*Engineering Thermodynamics* R. K. Singal, Mridul Singal, Rishi Singal, 2013-12-30 Engineering Thermodynamics has been designed for students of all branches of engineering specially undergraduate students of Mechanical Engineering The book will also serve as reference manual for practising engineers The book has been written in simple language and systematically develops the concepts and principles essential for understanding the subject The text has been supplemented with solved numerical problems illustrations and question banks The present book has been divided in five parts Thermodynamic Laws and Relations Properties of Gases and Vapours Thermodynamics Cycles Heat Transfer and Heat Exchangers Annexures

**Guide to RRB Junior Engineer Stage II Mechanical & Allied Engineering 3rd Edition** Disha Experts, 2019-03-02 Guide to RRB Junior Engineer Stage II Civil Allied Engineering 3rd Edition covers all the 5 sections including the Technical Ability Section in detail The book covers the complete syllabus as prescribed in the latest notification The book is divided into 5 sections which are further divided into chapters which contains theory explaining the concepts involved followed by Practice Exercises The Technical section is divided into 13 chapters The book provides the Past 2015 2014 Solved questions at the end of each section The book is also very useful for the Section Engineering Exam [Chemical Thermodynamics](#) Siddharth Venkatesh, 2025-02-20 Chemical Thermodynamics The Essentials offers a comprehensive and accessible exploration of the fundamental principles and practical applications of thermodynamics in chemical systems Designed for students researchers and professionals this book delves into the energetic underpinnings of chemical reactions and processes Covering basic principles to advanced topics like phase equilibria and chemical kinetics each chapter provides clear explanations illustrative examples and practical applications The book adopts a rigorous approach to ensure a solid understanding of the subject matter systematically presenting complex concepts and emphasizing a strong theoretical foundation Practical relevance is highlighted through applications in chemical engineering environmental science and materials science Thought provoking exercises accompany each chapter fostering critical thinking and practical problem solving Helpful pedagogical tools such as chapter summaries key terms and glossaries aid comprehension and serve as valuable references Beyond being a textbook Chemical Thermodynamics The Essentials aims to inspire curiosity and exploration in the field of thermodynamics Engaging narratives and insightful discussions encourage readers to delve deeper into the fascinating world of chemical energetics Whether you re a student or a seasoned researcher this book offers a comprehensive and engaging resource to deepen your understanding of chemical thermodynamics and unlock the mysteries of the energetic heart of chemistry [Fundamentals of Chemical Engineering Thermodynamics](#) Themis Matsoukas, 2012-10-02 The Clear Well Organized Introduction to Thermodynamics Theory and Calculations for All Chemical Engineering Undergraduate Students This text is designed to make thermodynamics far easier for undergraduate chemical engineering students to learn and to help them perform thermodynamic calculations with confidence Drawing on his award

winning courses at Penn State Dr Themis Matsoukas focuses on why as well as how He offers extensive imagery to help students conceptualize the equations illuminating thermodynamics with more than 100 figures as well as 190 examples from within and beyond chemical engineering Part I clearly introduces the laws of thermodynamics with applications to pure fluids Part II extends thermodynamics to mixtures emphasizing phase and chemical equilibrium Throughout Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering including separations reactions and capstone design More than 300 end of chapter problems range from basic calculations to realistic environmental applications these can be solved with any leading mathematical software Coverage includes Pure fluids PVT behavior and basic calculations of enthalpy and entropy Fundamental relationships and the calculation of properties from equations of state Thermodynamic analysis of chemical processes Phase diagrams of binary and simple ternary systems Thermodynamics of mixtures using equations of state Ideal and nonideal solutions Partial miscibility solubility of gases and solids osmotic processes Reaction equilibrium with applications to single and multiphase reactions

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## **Table of Contents Engineering Thermodynamics Question Bank**

1. Understanding the eBook Engineering Thermodynamics Question Bank
  - The Rise of Digital Reading Engineering Thermodynamics Question Bank
  - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Thermodynamics Question Bank
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Engineering Thermodynamics Question Bank
  - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Thermodynamics Question Bank
  - Personalized Recommendations
  - Engineering Thermodynamics Question Bank User Reviews and Ratings
  - Engineering Thermodynamics Question Bank and Bestseller Lists
5. Accessing Engineering Thermodynamics Question Bank Free and Paid eBooks
  - Engineering Thermodynamics Question Bank Public Domain eBooks
  - Engineering Thermodynamics Question Bank eBook Subscription Services
  - Engineering Thermodynamics Question Bank Budget-Friendly Options

6. Navigating Engineering Thermodynamics Question Bank eBook Formats
  - ePub, PDF, MOBI, and More
  - Engineering Thermodynamics Question Bank Compatibility with Devices
  - Engineering Thermodynamics Question Bank Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Engineering Thermodynamics Question Bank
  - Highlighting and Note-Taking Engineering Thermodynamics Question Bank
  - Interactive Elements Engineering Thermodynamics Question Bank
8. Staying Engaged with Engineering Thermodynamics Question Bank
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Engineering Thermodynamics Question Bank
9. Balancing eBooks and Physical Books Engineering Thermodynamics Question Bank
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Engineering Thermodynamics Question Bank
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Engineering Thermodynamics Question Bank
  - Setting Reading Goals Engineering Thermodynamics Question Bank
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Engineering Thermodynamics Question Bank
  - Fact-Checking eBook Content of Engineering Thermodynamics Question Bank
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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