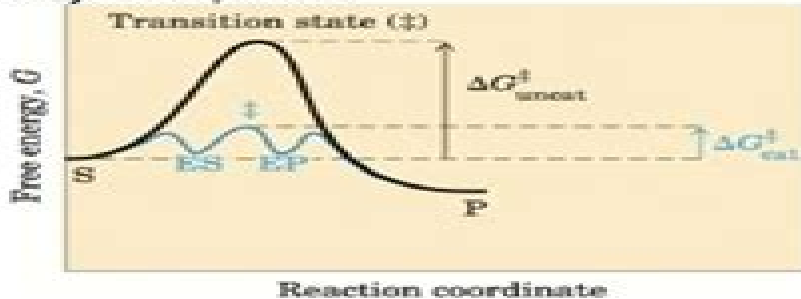


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ENZYMES PRINCIPLES, CLASSIFICATION, AND MECHANISM OF ACTION

ENZYMES

- Proteins that act as biological **catalysts**.
- At biological conditions, biomolecules are stable. For them to be metabolized, most reactions in the body rely on the action of enzymes to proceed.



ENZYME STRUCTURE

Adylyase (Adyly, $\text{C}_{12}\text{H}_{22}\text{O}_{11}$) – an enzyme found in saliva and pancreas – is responsible for the hydrolysis of sugars during digestion.

Active site (in red)

- the substrate binding and catalytic site of the enzyme.
- is complementary to the shape of the substrate.

Substrate (in yellow)

- the molecule which is acted upon by enzymes.
- In the case of Adylyase, the substrate (Adyly) is a short chain polysaccharide which will be broken down into smaller sugar units.

When the substrate is bound to the active site, the resulting structure is known as the **Enzyme-Substrate Complex**.

Coenzymes

- are substances which are organic molecules.
- multiple of coenzymes are the B vitamins.

Some enzymes need a coenzyme to function. The non-protein portion of the enzyme is known as an **Apoenzyme**.

Once a coenzyme is bound and the enzyme becomes active, it is known as a **Holoenzyme**.

Co-factors

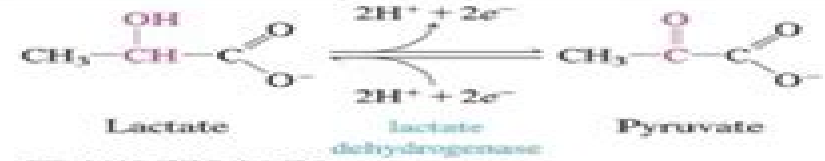
- are inorganic ions needed by some enzymes to be active.
- Adylyase needs Chloride (Cl^-) ions, generally used to cause an **hydrolysis** reaction.

CLASSIFICATION OF ENZYMES

- Enzymes are classified into six (6) major groups based on the reaction they catalyze.

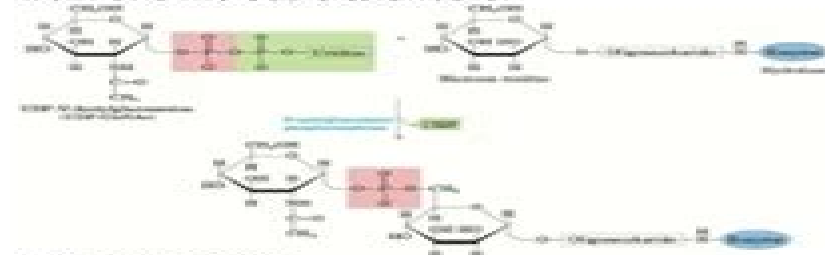
OXIDOREDUCTASES

- Catalyzes oxidation and reduction reactions.
- Usually requires a coenzyme to function (in the form of NADP^+ or NAD^+).



TRANSFERASES

- Catalyzes the transfer of a functional group from one molecule to another.



HYDROLASES

- Catalyze the removal of bonds via hydrolysis.



LYASES

- Catalyze the addition of groups to double bonds.
- Can also catalyze the formation of double bonds by removal of groups.



ISOMERASES

- Catalyze the interconversion of a molecule to its isomers.

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F. M. Strong



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Clinical Biochemistry Peter Rae, Mike Crane, Rebecca Pattenden, 2017-10-02 Clinical Biochemistry Lecture Notes presents the fundamental science behind common biochemical investigations used in clinical practice Taking a system based approach it explores the underlying physiological rationale for tests with each test explained within the context of disruption by disease It also explores the value and limitations of biochemical investigations while helping readers to quickly develop the knowledge and skills required to select the appropriate investigations for diagnosis and management and to correctly interpret test results Case studies throughout chapters place the information within a clinical context to further assist readers in the development of test selection and interpretation skills Key features include A comprehensive yet concise overview of the science behind common biochemical investigations Helps readers rapidly acquire a fully integrated practical understanding of biochemical diagnostics Full colour flowcharts and algorithms detailing the rationale for tests the biochemical processes involved and test procedures for quick comprehension and reference More clinical cases demonstrating application to practice Now in its tenth edition this classic introductory reference and revision text is indispensable to medical students and all those who want to quickly acquire a practical understanding of the scientific principles underpinning biochemical tests and a working knowledge of test selection test procedures and the interpretation of results within a clinical context

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