

Revision worksheet , Grade 11, Chemistry.

1. What is isomerism? Explain functional isomerism, metamerism and tautomerism with the help of suitable examples.
2. What do you mean by resonance. Explain with the help of an example.
3. What do you mean by Electron Displacement Effect? Explain inductive effect with the help of a proper example.
4. Write a note on the following reactions:-
 - a. Sabatier and Senderens reaction
 - b. Friedel Craft's reaction
 - c. Isomerisation
 - d. Dehydrohalogenation
 - e. Markonikov's rule
 - f. Ozonolysis
 - g. Decarboxylation
 - h. Wurtz reaction
 - i. Kolbe's electrolysis
 - j. Reforming or aromatization
 - k. Dehalogenation
 - l. Peroxide effect
 - m. Reaction of an alkene with KMnO_4
 - n. Dehydration
5. What is Grignard's reagent? How is it prepared? It was observed that Grignard's reagent should be stored in moisture proof containers. Why?
6. Identify the oxidizing agent, reducing agent, oxidant and reductant from the following using oxidation number method
$$\text{Zn} + 2\text{HCl} \longrightarrow \text{ZnCl}_2 + \text{H}_2$$
7. What is disproportionation reaction? Give an example.
8. What do you mean by an open system, closed system and an isolated system? Give an example.
9. What do you mean by saying that heat absorbed by a system is +ve and heat liberated by a system is -ve. Explain.
10. Explain the first law of Thermodynamics. Write its mathematical representation.
11. What do you mean by spontaneous and non spontaneous process. Give one example each.
12. State and explain Zeroth Law of Thermodynamics.
13. Explain Molarity, molality, mass percent and mole fraction.
14. Explain Aufbau principle, Hund's rule and Pauli's exclusion principle.
15. Draw the shapes of s,p and d orbitals.
16. What is ionization enthalpy? The ionization enthalpy of noble gases are very high. Why
17. What is electron gain enthalpy?

Solution Chemistry Grade 11

H Kauffman



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