

# Introduction to Solutions

**Instructions:** Circle the substances below that would be classified as **solutions**.

<p><b>Apple Juice</b></p> 	<p><b>Salad</b></p> 	<p><b>Dirt</b></p> 	<p><b>Lemonade</b></p> 
<p><b>Coffee with Sugar</b></p> 	<p><b>Pure Water (H<sub>2</sub>O)</b></p> 	<p><b>Cookie</b></p> 	<p><b>Soda Pop</b></p> 

**Instructions:** Use the choices in the word bank below to answer the following questions:

<b>Aqueous</b>	<b>Electrolyte</b>	<b>Solution</b>	<b>Water</b>	<b>Solute</b>	<b>Solvent</b>	<b>Solvation</b>
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- 1) Solute The substance that is dissolved in a solution (i.e. sugar in water)
- 2) Solution A homogenous mixture of two or more substances.
- 3) Electrolyte Solution that can conduct an electric current due to dissolved ions.
- 4) Solvation The process of dissolving. Solute particles are surrounded by solvent.
- 5) Aqueous A solution in which the solvent is water (H<sub>2</sub>O)
- 6) Solvent The substance that is doing the dissolving in a solution (i.e. water)
- 7) Water This substance is often referred to as the "universal solvent."

# Solutions Worksheets For Chemistry

**DP Hallahan**



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