

## Balancing Chemical Equations Worksheet

1.  $\underline{2} \text{ H}_2 + \underline{\quad} \text{ O}_2 \rightarrow \underline{2} \text{ H}_2\text{O}$
2.  $\underline{\quad} \text{ N}_2 + \underline{3} \text{ H}_2 \rightarrow \underline{2} \text{ NH}_3$
3.  $\underline{\quad} \text{ S}_8 + \underline{12} \text{ O}_2 \rightarrow \underline{8} \text{ SO}_3$
4.  $\underline{2} \text{ N}_2 + \underline{\quad} \text{ O}_2 \rightarrow \underline{2} \text{ N}_2\text{O}$
5.  $\underline{2} \text{ HgO} \rightarrow \underline{2} \text{ Hg} + \underline{\quad} \text{ O}_2$
6.  $\underline{4} \text{ CO}_2 + \underline{6} \text{ H}_2\text{O} \rightarrow \underline{\quad} \text{ C}_6\text{H}_{12}\text{O}_6 + \underline{6} \text{ O}_2$
7.  $\underline{\quad} \text{ Zn} + \underline{2} \text{ HCl} \rightarrow \underline{\quad} \text{ ZnCl}_2 + \underline{\quad} \text{ H}_2$
8.  $\underline{\quad} \text{ SiCl}_4 + \underline{4} \text{ H}_2\text{O} \rightarrow \underline{\quad} \text{ H}_4\text{SiO}_4 + \underline{4} \text{ HCl}$
9.  $\underline{2} \text{ Na} + \underline{2} \text{ H}_2\text{O} \rightarrow \underline{2} \text{ NaOH} + \underline{\quad} \text{ H}_2$
10.  $\underline{2} \text{ H}_3\text{PO}_4 \rightarrow \underline{\quad} \text{ H}_4\text{P}_2\text{O}_7 + \underline{\quad} \text{ H}_2\text{O}$
11.  $\underline{\quad} \text{ C}_{10}\text{H}_{16} + \underline{8} \text{ Cl}_2 \rightarrow \underline{10} \text{ C} + \underline{16} \text{ HCl}$
12.  $\underline{\quad} \text{ CO}_2 + \underline{2} \text{ NH}_3 \rightarrow \underline{\quad} \text{ OC}(\text{NH}_2)_2 + \underline{\quad} \text{ H}_2\text{O}$
13.  $\underline{4} \text{ Si}_2\text{H}_3 + \underline{17} \text{ O}_2 \rightarrow \underline{8} \text{ SiO}_2 + \underline{6} \text{ H}_2\text{O}_3$
14.  $\underline{2} \text{ Al}(\text{OH})_3 + \underline{3} \text{ H}_2\text{SO}_4 \rightarrow \underline{\quad} \text{ Al}_2(\text{SO}_4)_3 + \underline{6} \text{ H}_2\text{O}$
15.  $\underline{4} \text{ Fe} + \underline{3} \text{ O}_2 \rightarrow \underline{2} \text{ Fe}_2\text{O}_3$
16.  $\underline{\quad} \text{ Fe}_2(\text{SO}_4)_3 + \underline{6} \text{ KOH} \rightarrow \underline{3} \text{ K}_2\text{SO}_4 + \underline{2} \text{ Fe}(\text{OH})_3$
17.  $\underline{2} \text{ C}_7\text{H}_6\text{O}_2 + \underline{15} \text{ O}_2 \rightarrow \underline{14} \text{ CO}_2 + \underline{6} \text{ H}_2\text{O}$
18.  $\underline{\quad} \text{ H}_2\text{SO}_4 + \underline{8} \text{ HI} \rightarrow \underline{\quad} \text{ H}_2\text{S} + \underline{4} \text{ I}_2 + \underline{4} \text{ H}_2\text{O}$
19.  $\underline{4} \text{ FeS}_2 + \underline{11} \text{ O}_2 \rightarrow \underline{2} \text{ Fe}_2\text{O}_3 + \underline{8} \text{ SO}_2$
20.  $\underline{2} \text{ Al} + \underline{3} \text{ FeO} \rightarrow \underline{\quad} \text{ Al}_2\text{O}_3 + \underline{3} \text{ Fe}$
21.  $\underline{\quad} \text{ Fe}_2\text{O}_3 + \underline{3} \text{ H}_2 \rightarrow \underline{2} \text{ Fe} + \underline{3} \text{ H}_2\text{O}$
22.  $\underline{\quad} \text{ Na}_2\text{CO}_3 + \underline{2} \text{ HCl} \rightarrow \underline{2} \text{ NaCl} + \underline{\quad} \text{ H}_2\text{O} + \underline{\quad} \text{ CO}_2$
23.  $\underline{2} \text{ K} + \underline{\quad} \text{ Br}_2 \rightarrow \underline{2} \text{ KBr}$
24.  $\underline{\quad} \text{ C}_7\text{H}_{16} + \underline{11} \text{ O}_2 \rightarrow \underline{7} \text{ CO}_2 + \underline{8} \text{ H}_2\text{O}$
25.  $\underline{2} \text{ P}_4 + \underline{5} \text{ O}_2 \rightarrow \underline{2} \text{ P}_2\text{O}_5$

# Balancing Chemical Equations Answer Sheet

**F Rizvi**



## **Balancing Chemical Equations Answer Sheet:**

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**The Effectiveness of a  
Computer-assisted-instruction Package in Supplementing Teaching of Selected Concepts in High School  
Chemistry** Camille Linda Wainwright, 1984

This book delves into Balancing Chemical Equations Answer Sheet. Balancing Chemical Equations Answer Sheet is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Balancing Chemical Equations Answer Sheet, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Balancing Chemical Equations Answer Sheet
- Chapter 2: Essential Elements of Balancing Chemical Equations Answer Sheet
- Chapter 3: Balancing Chemical Equations Answer Sheet in Everyday Life
- Chapter 4: Balancing Chemical Equations Answer Sheet in Specific Contexts
- Chapter 5: Conclusion

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6. In chapter 5, the author will draw a conclusion about Balancing Chemical Equations Answer Sheet. The final chapter will summarize the key points that have been discussed throughout the book.

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