Name: Sam Li

Simulation: Ionic and Covalent Bonding

Background

In this investigation you will bond select atoms. Based upon the types of atoms that you choose to combine, you will create either an ionic compound or a covalent compound. You will have the opportunity to analyze the differences between these different types of compounds and to predict the number of atoms needed to create each, as well as learn how to appropriately name them.

1. Describe the difference between an atom and a molecule:

   Atoms are single particles, and molecules are made up of multiple atoms

2. Where are metal atoms located on the periodic table? Where are non-metal atoms located on the periodic table?

   Metals are on the left side and metals are on the right

3. What subatomic particle(s) participate in chemical bonding?

   electron

4. In your own words, define *valence electron*.

   electrons on outside ring which allow for bonding, negative charge

5. How can you determine the number of valence electrons in an atom using the periodic table?

   Look at the atomic #

6. Draw a Lewis Dot Structure for the following atoms:

a. Strontium (Sr)

b. Carbon (C)

c. Iodine (I)

d. Xenon (Xe)

*Check your answers before moving on to the next portion of the activity.

Ionic Bonding Lab Answers

Katrin Zwirglmaier

Ionic Bonding Lab Answers:

Lab Experiments in Introductory Chemistry Phil Reedy,Donald J. Wink,Sharon Fetzer-Gislason,2003-03-21 The manual contains laboratory experiments written specifically for the prep chem lab as well as for the general chemistry course Available as a complete manual or custom published athttp custompub whfreeman com **BIS Exam PDF-Technical Assistant (Lab) Chemical eBook PDF** Chandresh Agrawal,nandini books,2026-01-16 SGN The eBook BIS Technical Assistant Lab Chemical Covers Chemistry Subject Objective Questions From Various Exams With Answers **Lab Manual for General, Organic, and Biochemistry** Denise Guinn,Rebecca Brewer,2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www whfreeman com gob **Lab Manual Science Class 09** Neena Sinha, R.Rangarajan, Rajesh Kumar, These Lab Manuals provide complete information on all the experiments listed in the latest CBSE syllabus The various objectives materials required procedures inferences etc have been given in a step by step manner Carefully framed MCQs and short answers type questions given at the end of the experiments help the students prepare for viva voce Addison Wesley Chemistry 5th Edition Probeware Lab Manual 2002c Antony C. Wilbraham,Dennis D. Staley,Michael S. Matta,Edward L. Waterman,Prentice-Hall Staff,2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link **English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента** Надежда Зорина,Александра Соболева,2022-10-19 **PE Lab Exp(Noncons)Mod Chem 90** Tzimopoulo,1990

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you confidence as you embark on your career in science **A Laboratory Centered Molecular Biology Teaching Module which Facilitates the Teaching of Biological Science** Michael James Brundage,1990 **Environmental Health Perspectives** ,2000 **Instructors Manual to Lab Manual** Ralph Petrucci,William Harwood,Geoffrey Herring,2001 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Lab Exs In Prin Med Sci* Pearson Custom Publishing,1990 **Chemistry** Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02 *Addison-Wesley Small-scale Chemistry* Dennis D. Staley,Edward L. Waterman,1995 *Lab Experiments Modern Chemistry* Harold Clark Metcalfe,1982 *Biology/science Materials* Carolina Biological Supply Company,1991 Chemical Interactions ,2005 Focus on physical science Charles H. Heimler,1989 Steps to Doing Science Kristin Tuttle Bump,2008

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In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Ionic Bonding Lab Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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