

Structure of H₂O
hydrogen bonding
Denser at 4°C



unit one guide

Biological molecules
proteins
nucleic acids

H₂O provides cushion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, stable chemistry, low compressibility

hydrogen bonds require lots of energy to change state of matter

CONDENSATION - when two molecules combine, produce hydrogen and hydroxyl ions, releases H₂O

Hydroxyl - attracted to same surface
Hydrogen bond - attraction between hydrogen atoms of one water molecule

C	H	N	O	P	S
Carbon	Hydrogen	Nitrogen	Oxygen	Phosphorus	Sulfur

more hydrogen bonds
C - build biological molecules, cell formation
N - build proteins, nucleic acids
P - build nucleic acids + lipids
make up 98% of organisms

Carbohydrates ↓ C, H, O	Lipids ↓ C, H, O	Nucleic acids ↓ C, H, N, O, P	Proteins ↓ C, H, N, O
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Condensation synthesis - if the atoms of an H₂O molecule to bond molecules, molecule to release a bond between molecules (water, protein)

Nucleic acid structure:

5' end carbon sugar (deoxyribose/ribose), phosphate, nitrogen base (A, T, C, G), 3' end sugar of thymine



Starch vs Cellulose

Starch - alpha glucose
Cellulose - beta glucose

Proteins:

Specific order of amino acids
Primary structure

R group interactions determine structure + function

PROTEINS

amino acids bond via a peptide bond (covalent) through dehydration synthesis
R group change can cause a hydrophobic protein to lose the place of a hydrophilic protein

membranes

made up of phospholipids
create selectively permeable barrier
contain phospholipids, cholesterol, and proteins



LIPIDS

Phospholipid macromolecules

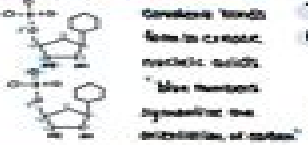
differentiable in saturation determine structure and function
Cells, sterols, phospholipids
Saturated - no double bonds
Unsaturated - has double bonds
4 in 1: triglycerides, cholesterol, make up the membrane



Amino Acids:

linear chains, connected by covalent bonds
amino end and carboxyl end

DEHYDRATION SYNTHESIS



covalent bonds form to create nucleic acids
the number specifies the structure of water

PROTEIN STRUCTURE

Primary: sequence of amino acids, specific order determine protein shape
Secondary: how folding of chain into alpha helix and beta sheet occurs
Tertiary: 3-D shape, hydrogen bonds, disulfide link, ionic bonds, hydrophobic interactions
Quaternary: interactions between multiple polypeptide chains

+ FUNCTION

Enzymes - speed up reaction rates in the cell
Cell structure - part of the cellular membrane
Communication - between cells

DENATURATION:

the protein starts to lose its shape/structure when (bonds break)
temperature change
pH of surrounding area
salt concentration

NH₂ = Amino
COOH = Carboxyl

NUCLEIC ACIDS

add new nucleotides to the 3' end
two nitrogenous bases: purine + pyrimidine
pyrimidine is single ringed, purine is double
antiparallel double helix
5' on phosphate end, 3' on hydroxyl end
both back and forth have nitrogenous bases (purines) + sugar-phosphate backbone

ANTIPARALLEL



Chaperones:

protein molecules that assist protein folding

ALDOSE VS KETOSE

contains aldehyde functional group

ALPHA VS BETA

contains ketone functional group



ALPHA VS BETA

all monomers are in the same orientation
Every glucose monomer is upside down from its neighbor

Ap Biology Chapter 12 Notes

Shasha Hu



Ap Biology Chapter 12 Notes:

Biomacromolecules Mohamed Elzagheid,2024-12-30 The second edition of the Macromolecular Chemistry broadens into two areas biomacromolecules Volume 1 and polymers Volume 2 Biomacromolecules covers carbohydrates lipids proteins nucleic acids their classifications and properties The first chapter looks at the structural formulas and cyclic forms of monosaccharides as well as their synthesis and breakdown Cyclization enolization isomerization tautomerization mutarotation and epimerization are all briefly described Examples of disaccharides and polysaccharides are also presented The second chapter covers triglycerides steroids vitamins and their constituents The third chapter examines the primary structure of proteins including amino acid properties peptide bond formation and peptide synthesis It also addresses secondary tertiary and quaternary structures The book concludes with a chapter on nucleic acids which covers the chemistry of nucleosides and oligonucleotides as well as topics such as genetic code DNA secret code Polymerase Chain Reaction and DNA fingerprinting

Game of Fear: A gripping psychological thriller Gledé Browne Kabongo,2016-02-24 Sometimes good girls do very bad things Academic overachiever and popular good girl Abbie Cooper is used to being in control and strives to keep it that way As senior year at her elite New England private school kicks off she s focused on one goal getting into the Ivy League But even good girls have dark secrets One morning Abbie discovers a cryptic note in her locker I know what you did hypocrite Justice will be served The Avenger Then a photo arrives in the mail It captures in great detail the explosive secret she s kept hidden from everyone the shocking blunder that could get her expelled from Saint Matthews Academy and keep her out of the Ivy League for good Determined to protect her future and the people she loves Abbie is drawn into a diabolical game of blackmail cyber spying and deception For 50 000 the mystery girl known as The Avenger assures Abbie that her secret will remain safe There s only one problem The Avenger may not exist at all So who s stalking Abbie with anonymous text messages that disappear after they ve been read Who s out to ruin her and why To uncover the horrifying truth Abbie must play the game all the way to its stunning conclusion Provided by publisher

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Fulder,1980 **Latin Notes** ,1926 *New Technical Books* ,1929 Ladies in the Laboratory? American and British Women in Science, 1800-1900 Mary R.S. Creese,2000-01-01 A systematic survey and comparison of the work of 19th century American and British women in scientific research this book covers the two countries in which women of the period were most active in scientific work and examines all the fields in which they were engaged The field by field examination brings out patterns and concentrations in women s research in both countries and allows a systematic comparison of the two national groups Through this comparison new insights are provided into how the national patterns developed and what they meant in terms of both the process of women s entry into research and the contributions they made there Ladies in the Laboratory features a specialized bibliography of nineteenth century research journal publications by women created from the London Royal Society s Catalogue of Scientific Papers 1800 1900 In addition 23 illustrations present in condensed form information about American and British women s scientific publications throughout the nineteenth century This well organized blend of individual life stories and quantitative information presents a great deal of new data and field by field analysis its broad and methodical coverage will make it a basic work for everyone interested in the story of women s participation in nineteenth century science The Ohio Journal of Science ,1921 Includes book reviews and abstracts

Biological Science Biological Sciences Curriculum Study,1963 *International Index to Periodicals* ,1924 An author and subject index to publications in fields of anthropology archaeology and classical studies economics folklore geography history language and literature music philosophy political science religion and theology sociology and theatre arts *The Literary World* ,1891 **Biological & Agricultural Index** ,1928 **Reader's Guide to Periodical Literature Supplement** ,1923

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