

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 18. A column in the periodic table | _____ a. metals |
| _____ 19. A row in the periodic table | _____ b. group |
| _____ 20. Group 1 elements | _____ c. period |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are (or) noble gases elements known as inert gases.
- _____ 29. Alkaloids have properties of both metals and more reactive metals.

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G Psacharopoulos



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Table of Contents Chapter 11 Study Guide Content Mastery Answers

1. Understanding the eBook Chapter 11 Study Guide Content Mastery Answers
 - The Rise of Digital Reading Chapter 11 Study Guide Content Mastery Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 11 Study Guide Content Mastery Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 11 Study Guide Content Mastery Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 11 Study Guide Content Mastery Answers
 - Personalized Recommendations
 - Chapter 11 Study Guide Content Mastery Answers User Reviews and Ratings
 - Chapter 11 Study Guide Content Mastery Answers and Bestseller Lists

5. Accessing Chapter 11 Study Guide Content Mastery Answers Free and Paid eBooks
 - Chapter 11 Study Guide Content Mastery Answers Public Domain eBooks
 - Chapter 11 Study Guide Content Mastery Answers eBook Subscription Services
 - Chapter 11 Study Guide Content Mastery Answers Budget-Friendly Options
6. Navigating Chapter 11 Study Guide Content Mastery Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 11 Study Guide Content Mastery Answers Compatibility with Devices
 - Chapter 11 Study Guide Content Mastery Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 11 Study Guide Content Mastery Answers
 - Highlighting and Note-Taking Chapter 11 Study Guide Content Mastery Answers
 - Interactive Elements Chapter 11 Study Guide Content Mastery Answers
8. Staying Engaged with Chapter 11 Study Guide Content Mastery Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 11 Study Guide Content Mastery Answers
9. Balancing eBooks and Physical Books Chapter 11 Study Guide Content Mastery Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 11 Study Guide Content Mastery Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 11 Study Guide Content Mastery Answers
 - Setting Reading Goals Chapter 11 Study Guide Content Mastery Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 11 Study Guide Content Mastery Answers
 - Fact-Checking eBook Content of Chapter 11 Study Guide Content Mastery Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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