

Chapter 12

SECTION 12.1

12.1.1 Boyle's Law: Pressure and Volume of a Gas at Constant Temperature

1. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume at constant temperature. Boyle's Law can be expressed as follows:

Boyle's Law: $P_1V_1 = P_2V_2$ (at constant temperature and mass)

Boyle's Law: $P \propto \frac{1}{V}$ (at constant temperature and mass)

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2. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume at constant temperature. Boyle's Law can be expressed as follows:

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3. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume at constant temperature. Boyle's Law can be expressed as follows:

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7. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume at constant temperature. Boyle's Law can be expressed as follows:

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