

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	56	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

Chapter 12 Basic Review Worksheet Chemistry Answers

Chao Zhang



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Table of Contents Chapter 12 Basic Review Worksheet Chemistry Answers

1. Understanding the eBook Chapter 12 Basic Review Worksheet Chemistry Answers
 - The Rise of Digital Reading Chapter 12 Basic Review Worksheet Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 12 Basic Review Worksheet Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 12 Basic Review Worksheet Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 12 Basic Review Worksheet Chemistry Answers
 - Personalized Recommendations
 - Chapter 12 Basic Review Worksheet Chemistry Answers User Reviews and Ratings
 - Chapter 12 Basic Review Worksheet Chemistry Answers and Bestseller Lists
5. Accessing Chapter 12 Basic Review Worksheet Chemistry Answers Free and Paid eBooks
 - Chapter 12 Basic Review Worksheet Chemistry Answers Public Domain eBooks
 - Chapter 12 Basic Review Worksheet Chemistry Answers eBook Subscription Services

- Chapter 12 Basic Review Worksheet Chemistry Answers Budget-Friendly Options
- 6. Navigating Chapter 12 Basic Review Worksheet Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 12 Basic Review Worksheet Chemistry Answers Compatibility with Devices
 - Chapter 12 Basic Review Worksheet Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 12 Basic Review Worksheet Chemistry Answers
 - Highlighting and Note-Taking Chapter 12 Basic Review Worksheet Chemistry Answers
 - Interactive Elements Chapter 12 Basic Review Worksheet Chemistry Answers
- 8. Staying Engaged with Chapter 12 Basic Review Worksheet Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 12 Basic Review Worksheet Chemistry Answers
- 9. Balancing eBooks and Physical Books Chapter 12 Basic Review Worksheet Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 12 Basic Review Worksheet Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 12 Basic Review Worksheet Chemistry Answers
 - Setting Reading Goals Chapter 12 Basic Review Worksheet Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 12 Basic Review Worksheet Chemistry Answers
 - Fact-Checking eBook Content of Chapter 12 Basic Review Worksheet Chemistry Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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