

4. A solution contains 2.0 g of FeSO_4 in 100 mL of solution. What is the molarity of the solution? (Molar mass of FeSO_4 is 152 g/mol.)
- 0.013 M
 - 0.013 g
 - 0.013 mol
 - 0.013 L
5. A solution contains 2.0 g of FeSO_4 in 100 mL of solution. What is the molarity of the solution? (Molar mass of FeSO_4 is 152 g/mol.)
- 0.013 M
 - 0.013 g
 - 0.013 mol
 - 0.013 L
6. The molarity of a solution is 0.5 M. What is the mass of the solute in 100 mL of the solution? (Molar mass of the solute is 100 g/mol.)
- 5.0 g
 - 50.0 g
 - 0.5 g
 - 0.50 g
7. A solution contains 2.0 g of FeSO_4 in 100 mL of solution. What is the molarity of the solution? (Molar mass of FeSO_4 is 152 g/mol.)
- 0.013 M
 - 0.013 g
 - 0.013 mol
 - 0.013 L
8. The molarity of a solution is 0.5 M. What is the mass of the solute in 100 mL of the solution? (Molar mass of the solute is 100 g/mol.)
- 5.0 g
 - 50.0 g
 - 0.5 g
 - 0.50 g
9. A solution contains 2.0 g of FeSO_4 in 100 mL of solution. What is the molarity of the solution? (Molar mass of FeSO_4 is 152 g/mol.)
- 0.013 M
 - 0.013 g
 - 0.013 mol
 - 0.013 L
10. The molarity of a solution is 0.5 M. What is the mass of the solute in 100 mL of the solution? (Molar mass of the solute is 100 g/mol.)
- 5.0 g
 - 50.0 g
 - 0.5 g
 - 0.50 g
11. A solution contains 2.0 g of FeSO_4 in 100 mL of solution. What is the molarity of the solution? (Molar mass of FeSO_4 is 152 g/mol.)
- 0.013 M
 - 0.013 g
 - 0.013 mol
 - 0.013 L
12. The molarity of a solution is 0.5 M. What is the mass of the solute in 100 mL of the solution? (Molar mass of the solute is 100 g/mol.)
- 5.0 g
 - 50.0 g
 - 0.5 g
 - 0.50 g

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