

# 17

## WATER AND AQUEOUS SYSTEMS

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### SECTION 17.1 LIQUID WATER AND ITS PROPERTIES (pages 475–478)

*This section explains the hydrogen bonding that occurs in water molecules and how high surface tension and low vapor pressure can be explained in terms of hydrogen bonding.*

#### ► The Water Molecule (pages 475–476)

- What unique molecular compound is the foundation of all life on Earth?  
water
- Approximately what fraction of Earth's surface is covered in water? 3/4
- Circle the letter next to each sentence that is true concerning water molecules.
  - Each O — H covalent bond in a water molecule is nonpolar.
  - b. In a water molecule, the less electronegative hydrogen atoms acquire a slight positive charge and the oxygen atom acquires a slight negative charge.
  - c. Because the water molecule has an H — O — H bond angle of 105°, the molecule as a whole is polar.
- The diagram below depicts a water molecule. Complete the labels showing the locations of the hydrogen atoms, the oxygen atom, and the regions of positive and negative charge.



- List five important properties of water that result from hydrogen bonding.
  - high surface tension
  - low vapor pressure
  - high specific heat capacity
  - high heat of vaporization
  - high boiling point

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# Chapter 15 Water And Aqueous Systems Workbook Answers

**S Ashworth**



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Reversible Polymeric Gels and Related Systems Paul S. Russo,1987      High School Chemistry Teachers Magazine ,1967

**Chemical Modeling in Aqueous Systems** Everett A. Jenne,American Chemical Society. Division of Environmental Chemistry,1979      **Geochemistry** William M. White,2020-10-02 A Comprehensive Introduction to the Geochemist Toolbox the Basic Principles of Modern Geochemistry In the new edition of William M White s Geochemistry undergraduate and graduate students will find each of the core principles of geochemistry covered From defining key principles and methods to examining Earth s core composition and exploring organic chemistry and fossil fuels this definitive edition encompasses all the information needed for a solid foundation in the earth sciences for beginners and beyond For researchers and applied scientists this book will act as a useful reference on fundamental theories of geochemistry applications and environmental sciences The new edition includes new chapters on the geochemistry of the Earth s surface the critical zone marine geochemistry and applied geochemistry as it relates to environmental applications and geochemical exploration A review of the fundamentals of geochemical thermodynamics and kinetics trace element and organic geochemistry An introduction to radiogenic and stable isotope geochemistry and applications such as geologic time ancient climates and diets of prehistoric people Formation of the Earth and composition and origins of the core the mantle and the crust New chapters that cover soils and streams the oceans and geochemistry applied to the environment and mineral exploration In this foundational look at geochemistry new learners and professionals will find the answer to the essential principles and techniques of the science behind the Earth and its environs      Cellulose Chemistry and Technology ,1989      *Geochimica Et Cosmochimica Acta*

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*Selected Water Resources Abstracts* ,1986      **Annual Review of Physical Chemistry** Gerhard Krohn Rollefson,1956 Provides abstracts and review articles on topics in physical chemistry      **The Journal of Physical Chemistry** Wilder Dwight Bancroft,1919 Includes section New Books      *The Journal of Physical Chemistry* ,1919      Walford's Guide to Reference Material: Science and technology Marilyn Mullan,Library Association,1989 The Brit counterpart to Sheehy in which it is recommended The new edition places the author title subject indices in each volume Many entries cite reviews from other sources Rather tiresome recitation of selected chapter contents 6 000 entries with references in the annotations

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