

Test Bank

For

**STATISTICS:
THE ART AND SCIENCE
OF LEARNING FROM
DATA
THIRD EDITION**

**Alan Agresti
Christine Franklin**

Answers At The End Of Each Chapter

Statistics The Art And Science Of Learning From Data
3rd Edition

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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others

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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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nitrite perphosphate hypoiodite chlorite phosphite percarbonate bromate hyposulfite permanganate carbonite CO_3^{2-} MnO_3^- SO_5^{2-} BrO_2^- CO_4^{2-} PO_5^{3-} ClO_1^- IO_2^- PO_4^{3-} NO_2^- SO_4^{2-}

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web polyatomic ions are charged groups of atoms an example is ammonium ion NH_4^+ it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is NH_4Cl and ammonium sulfide is NH_4S

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion NO_3^- contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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web aug 11 2022 to identify and name polyatomic ions some ions consist of groups of atoms covalently bonded together and have an overall electric charge because these ions contain more than one atom they are called polyatomic ions the structures

names and formulas of some polyatomic ions are found in the below figure and table

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written

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