

CHAPTER 8 REVIEW

Chemical Equations and Reactions

SHORT ANSWER Answer the following questions in the space provided.

2. _____ According to the law of conservation of mass, the total mass of the reacting substances is
- (a) always more than the total mass of the products.
 - (b) always less than the total mass of the products.
 - (c) sometimes more and sometimes less than the total mass of the products.
 - (d) always equal to the total mass of the products.**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 2. In the reaction described by the word equation
- $$\text{sodium} + \text{water} \rightarrow \text{sodium hydroxide} + \text{hydrogen gas}$$

- a. sodium hydroxide is a product.
- b. sodium is a product.
- c. water is a reactant.
- d. Both (a) and (c).**

- _____ 3. Which of the following indicates that a chemical equation is balanced?

- a. The numbers of atoms of each element are the same on both sides of the equation.**
- b. All of the coefficients are the same.
- c. The numbers of molecules on each side of the equation are equal.
- d. The sums of the coefficients on each side of the equation are equal.

- _____ 4. From a complete and correctly written chemical equation, you can obtain the

- a. chemical formulas of the reactants and products.
- b. relative amounts of the reactants and products.
- c. physical states of the reactants and products.
- d. All of the above.**

- _____ 5. To balance a chemical equation, you must

- a. adjust the subscripts.
- b. adjust the number of products.
- c. adjust the number of reactants.
- d. adjust the coefficients.**

- _____ 6. The formation of a solid product in a chemical reaction is represented by the symbol

- a. (aq).
- b. (g).
- c. (l).
- d. (s).**

Chapter 8 Test Chemical Equations And Reactions Modern Chemistry

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Barnes, Ray Doerhoff, Max Bolen**



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Basic Chemical Principles
Donada Peters, Edward I. Peters, 1988 *The Algebra of Organic Synthesis* John Andraos, 2016-04-19 The Algebra of Organic Synthesis combines the aims philosophies and efforts involved in organic synthesis reaction optimization and green chemistry with techniques for determining quantitatively just how green synthesis plans are It provides the first complete quantitative description of synthesis strategy analysis in the context of green ch **Modern Chemical Technology** Chemical Technician Curriculum Project. Writing Team, 1972 *Applied Chemoinformatics* Thomas Engel, Johann Gasteiger, 2018-06-05 Edited by world famous pioneers in chemoinformatics this is a clearly structured and applications oriented approach to the topic providing up to date and focused information on the wide range of applications in this exciting field The authors explain methods and software tools such that the reader will not only learn the basics but also how to use the different software packages available Experts describe applications in such different fields as structure spectra correlations virtual screening prediction of active sites library design the prediction of the properties of chemicals the development of new cosmetics products quality control in food the design of new materials with improved properties toxicity modeling assessment of the risk of chemicals and the control of chemical processes The book is aimed at advanced students as well as lectures but also at scientists that want to learn how chemoinformatics could assist them in solving their daily scientific tasks Together with the corresponding textbook Chemoinformatics Basic Concepts and Methods ISBN 9783527331093 on the fundamentals of chemoinformatics readers will have a comprehensive overview of the field **Heath Science Connections 9** Jack L. Candido, Edward S. James, Ronald E. Phillips, 1987 High School Chemistry Teachers Magazine, 1967 **Chemical Engineering Catalog**, 1919 Econometrics: Alchemy Or Science? David F. Hendry, 2000-10-26 Econometrics Alchemy or Science analyses the effectiveness and validity of applying econometric methods to economic time series The methodological dispute is long standing and no claim can be made for a single valid method but recent results on the theory and practice of model selection bid fair to resolve many of the contentious issues The book presents criticisms and evaluations of competing approaches based on theoretical economic and econometric analyses empirical applications and Monte Carlo simulations which interact to determine best practice It explains the evolution of an approach to econometric modelling founded in careful statistical analyses of the available data using economic theory to guide the general model specification From a strong foundation in the theory of reduction via a range of applied and simulation studies it demonstrates that general to specific procedures have excellent properties The book is divided into four Parts Routes and Route Maps Empirical Modelling Strategies Formalization and Retrospect and Prospect A short preamble to each chapter sketches the salient themes links to earlier and later developments and the lessons learnt or missed at the time A sequence of detailed empirical studies of consumers expenditure and money demand illustrate most

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