

Chapter Test A

Chapter: Chemical Equations and Reactions

As this space grid fills, so will all the other spaces on pages that have completed with content on these answers each question.

1. Which statement is true, according to us in nature? If both of the following statements show that both are true, choose the one that is more appropriate.
 - a. The number of molecules is constant.
 - b. The number of molecules is not constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.
2. In the reaction described by the chemical equation:

$$\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$
 which of the following is true?
 - a. The number of molecules is constant.
 - b. The number of molecules is not constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.
3. Which of the following statements is true, according to us in nature? If both of the following statements show that both are true, choose the one that is more appropriate.
 - a. The number of molecules is constant, but the number of atoms is not constant.
 - b. The number of molecules is not constant, but the number of atoms is constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.
4. Which of the following statements is true, according to us in nature? If both of the following statements show that both are true, choose the one that is more appropriate.
 - a. The number of molecules is constant, but the number of atoms is not constant.
 - b. The number of molecules is not constant, but the number of atoms is constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.
5. Which of the following statements is true, according to us in nature? If both of the following statements show that both are true, choose the one that is more appropriate.
 - a. The number of molecules is constant, but the number of atoms is not constant.
 - b. The number of molecules is not constant, but the number of atoms is constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.
6. Which of the following statements is true, according to us in nature? If both of the following statements show that both are true, choose the one that is more appropriate.
 - a. The number of molecules is constant, but the number of atoms is not constant.
 - b. The number of molecules is not constant, but the number of atoms is constant.
 - c. The number of molecules is not constant, but the number of atoms is constant.
 - d. The number of molecules is not constant, but the number of atoms is constant.

Chapter 9 Modern Chemistry Test

JG Myers



Chapter 9 Modern Chemistry Test:

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