

$$\text{a) Rate} = \frac{-1}{2} \frac{d[\text{NO}]}{dt} = \frac{-d[\text{O}_2]}{dt} = \frac{1}{2} \frac{d[\text{NO}_2]}{dt}$$

$$\text{b) } \frac{-d[\text{O}_2]}{dt} = \frac{1}{2} \frac{d[\text{NO}_2]}{dt}$$

$$\frac{d[\text{NO}_2]}{dt} = 2 \times \left(\frac{-d[\text{O}_2]}{dt} \right) = 2 \times 0.2 \text{ mol L}^{-1} \text{ s}^{-1}$$

$$= 0.4 \text{ mol L}^{-1} \text{ s}^{-1}$$

Chemistry Chemical Kinetics Problems Answers

Jicheng Xie



Chemistry Chemical Kinetics Problems Answers:

Solutions Manual to Accompany Elements of Physical Chemistry David Smith, 2013-05-30 The Solutions Manual to accompany Elements of Physical Chemistry 6th edition contains full worked solutions to all end of chapter discussion questions and exercises featured in the book The manual provides helpful comments and friendly advice to aid understanding It is also a valuable resource for any lecturer who wishes to use the extensive selection of exercises featured in the text to support either formative or summative assessment and wants labour saving ready access to the full solutions to these questions

Cracking the GRE Chemistry Subject Test Princeton Review (Firm), 2005 The GRE subject tests are among the most difficult standardized exams Rather than testing general problem solving skills they require highly specialized knowledge The experts at The Princeton Review have thoroughly research each subject test to provide students with the most thorough up to date information available Students don t need to relearn the entire histories of their fields just what they need to know to earn high scores on the exams Each guide includes one full length practice exam complete with comprehensive explanations for every solution

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