

## Chapter Test II

### A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

#### Column I

1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
2. An oxidation-reduction reaction in which a substance is oxidized and the same substance is reduced.
3. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
4. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
5. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.
6. An oxidation-reduction reaction in which a substance is oxidized and another substance is reduced.

#### Column II

- a. oxidation-reduction
- b. redox reaction
- c. disproportionation
- d. combustion
- e. synthesis
- f. decomposition
- g. single displacement
- h. double displacement

### B. Multiple Choice

Choose the best answer for each question.

1. Which of the following is a redox reaction?
  - a.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
  - b.  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
  - c.  $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$
  - d.  $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$
2. Which of the following is a redox reaction?
  - a.  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
  - b.  $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$
  - c.  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
  - d.  $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$
3. Which of the following is a redox reaction?
  - a.  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
  - b.  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
  - c.  $\text{NaCl} + \text{KBr} \rightarrow \text{NaBr} + \text{KCl}$
  - d.  $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$

# Chemistry Chapter 12 Stoichiometry Test

**Richard H. Langley, John Moore**



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