

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) mass and volume. A(2) substance is a form of matter with a uniform and unchanging composition.Substances have specific, unchanging (3) properties that can be observed.Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons gain cohesion when exposed to heat or
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Crystals
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is $100^\circ C$
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{2} \frac{g}{cm^3}$

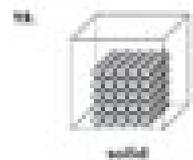
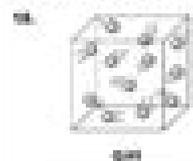
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CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 **continued**

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of three physical states.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

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Minjie Lin



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Table of Contents Chemistry Matter Change Chapter 12 Answers

1. Understanding the eBook Chemistry Matter Change Chapter 12 Answers
 - The Rise of Digital Reading Chemistry Matter Change Chapter 12 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Matter Change Chapter 12 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Chemistry Matter Change Chapter 12 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Matter Change Chapter 12 Answers
 - Personalized Recommendations
 - Chemistry Matter Change Chapter 12 Answers User Reviews and Ratings
 - Chemistry Matter Change Chapter 12 Answers and Bestseller Lists
5. Accessing Chemistry Matter Change Chapter 12 Answers Free and Paid eBooks
 - Chemistry Matter Change Chapter 12 Answers Public Domain eBooks
 - Chemistry Matter Change Chapter 12 Answers eBook Subscription Services
 - Chemistry Matter Change Chapter 12 Answers Budget-Friendly Options
6. Navigating Chemistry Matter Change Chapter 12 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Matter Change Chapter 12 Answers Compatibility with Devices
 - Chemistry Matter Change Chapter 12 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Matter Change Chapter 12 Answers
 - Highlighting and Note-Taking Chemistry Matter Change Chapter 12 Answers
 - Interactive Elements Chemistry Matter Change Chapter 12 Answers

8. Staying Engaged with Chemistry Matter Change Chapter 12 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Matter Change Chapter 12 Answers
9. Balancing eBooks and Physical Books Chemistry Matter Change Chapter 12 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Matter Change Chapter 12 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Matter Change Chapter 12 Answers
 - Setting Reading Goals Chemistry Matter Change Chapter 12 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Matter Change Chapter 12 Answers
 - Fact-Checking eBook Content of Chemistry Matter Change Chapter 12 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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