

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative  
positive  
crisscross

noble gas  
electrons  
anion

octet  
ions  
cation

electron configuration  
polyatomic  
empirical formula

- An ionic bond is an attraction between oppositely charged ions.
- Anions have a negative charge.
- An atom becomes an ion by losing or gaining electrons.
- The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
- The atoms that make up polyatomic ions are bonded together by covalent bonds.
- The crisscross method can be used to write the formula for an ionic compound.
- The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

- |                        |  |
|------------------------|--|
| <u>losing</u>          | 9. Calcium becomes a monatomic cation by <u>gaining</u> two electrons.                     |
| <u>true</u>            | 10. A cation has a <u>positive</u> charge.   |
| <u>true</u>            | 11. A binary ionic compound contains only <u>one</u> kind of cation and one kind of anion. |
| <u>polyatomic</u>      | 12. <u>Monatomic</u> ions consist of more than one atom.                                   |
| <u>seven</u>           | 13. The Lewis dot diagram for chlorine, a group 7A element, has <u>six</u> electrons.      |
| <u>true (released)</u> | 14. A <u>great deal of energy is produced</u> when an ionic compound is formed.            |

# Chemistry Review And Reinforcement Answers

**DP Hallahan**



## **Chemistry Review And Reinforcement Answers:**

## Unveiling the Magic of Words: A Overview of "**Chemistry Review And Reinforcement Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Chemistry Review And Reinforcement Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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