

# CUET Chemistry MCQs

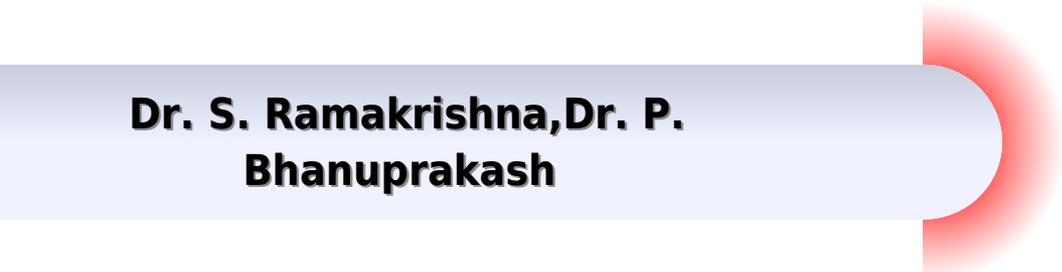
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## Solutions

1. Which of the following is not a binary solution?  
(A) Pure water + Sugar  
(B) Air  
(C) Mixture of benzene and toluene  
(D) Mixture of ethanol and methanol
2. Which of the following is not an example of saturated solution?  
(A) Sugar syrup  
(B) Salt water  
(C) Honey  
(D) Carbonated drink
3. If sea water is assumed to contain 40g of salt per 200g of sea water then percentage by mass of salt present is  
(A) 15%  
(B) 40%  
(C) 20%  
(D) 10%
4. If 2 moles of  $\text{CaCO}_3$  is dissolved in 900g of water then percentage by mass of solution formed is  
(A) 36.36%  
(B) 9.9%  
(C) 18.18%  
(D) 11.75%
5. 0.5 moles of benzene was dissolved in 2 moles of toluene. Find mass % of benzene in the solution formed  
(A) 25.48%  
(B) 9.52%  
(C) 15.92%  
(D) 17.48%
6. What is the number of ppm of NaCl in the solution with 117g of NaCl dissolved in 500 mL of water?  
(A)  $1.89 \times 10^5$   
(B)  $2.78 \times 10^5$   
(C)  $9.12 \times 10^5$   
(D)  $6.54 \times 10^5$
7. What is the concentration of mercury in ppm if 3g mercury is found in 15kg sample of ocean water?  
(A) 300  
(B) 400  
(C) 500  
(D) 200
8. If 3.0 mg of  $\text{Cr}^{+6}$  is dissolved in 3 L of water, find the concentration of  $\text{Cr}^{+6}$  in ppm.  
(A) 2 ppm  
(B) 1 ppm  
(C) 3 ppm  
(D) 9 ppm
9. What is the number of ppm of methanol dissolved in 68g of water (Taking mass of methanol 32gm)  
(A) 160000  
(B) 480000  
(C) 320000  
(D) 520000
10. A tank contains 5 moles of oxygen, 2 moles of nitrogen and 20g of hydrogen at room temperature. Find the mole fraction of hydrogen.  
(A)  $\frac{5}{17}$   
(B)  $\frac{2}{15}$   
(C)  $\frac{8}{15}$   
(D)  $\frac{10}{17}$
11. Which of the following is insoluble in water?  
(A) salt  
(B) acid  
(C) base  
(D) oil
12. On increasing temperature, the solubility of NaOH in water.  
(A) Increases  
(B) Decreases  
(C) Remains unaffected  
(D) Cannot be determined
13. On increasing temperature, the solubility of  $\text{NaNO}_3$  in water.  
(A) Increases  
(B) Decreases  
(C) Remains unaffected  
(D) Cannot be determined
14. What is the effect of complex formation on solubility of  $\text{I}_2$ ?  
(A) Increases  
(B) Decreases  
(C) Remains same  
(D) None of the above
15. Solid forms of carboxylic acids are soluble in  
(A) Water  
(B) Acetone  
(C) Wax  
(D) Oil
16. Fishes feel uncomfortable in warm water due to  
(A) fishes do not like warmth  
(B) higher amount of impurities

# Chemistry Solutions Questions And Answers

**Dr. S. Ramakrishna, Dr. P.  
Bhanuprakash**



## **Chemistry Solutions Questions And Answers:**

Physical Chemistry (Solutions, Photochemistry, Electrochemistry) Dr. S. Ramakrishna, Dr. P. Bhanuprakash, 2024-08-01

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**Chemistry** Phillip John Buddle, J. S. Mackay, 1970 *School Science and Mathematics*, 1919

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