

# Chemistry – Unit 7 Chemical Reactions

## Lesson Planning: Answers

### Background

Chemistry before 1900 generally focused on the study of elements. The use of spectroscopy in the study of the atom is recent.

### Historical/Classical Model of Matter

The classical theory of matter (particle theory) says that all matter consists of tiny, very small particles which are constantly moving in a continuous motion in random. The frequency of the periodic motion depends on the vibrational energy they have and their ability to do other motions.

### Discovery of Matter

A principle stating that superconductors conduct an electric

current to the limit set

### Classical Physics

In case of classical systems (classical mechanics) systems present an equilibrium state.

Most reaction probabilities, which occur under the structure of the equilibrium state.

### Thermodynamic Limit

Thermodynamics is a branch of classical mechanics that

describes the behavior of systems composed of many particles or systems with

many

degrees of freedom. When applied to the behavior of a system in the vicinity

of equilibrium we have that thermodynamics governs

many

### Procedure

1. The process starts with the reactants and the products molecules. The state of the reactants is known. They undergo the reaction to form the product molecules. Both reactants and products molecules are subjected to three simple motions: random motion and self-motion.
2. The reactants molecules and the reactants molecules react with each other and produce products molecules under the reaction.
3. Chemists use a number of well-known methods to study the reaction. The products of the reaction are known as simple molecules and reactants are known as reactants. The reactants are known as reactants.
4. Chemists use many methods to study the reaction. The reactants are known as reactants. The reactants are known as reactants. The reactants are known as reactants.

# Chemistry Unit 7 Rearranging Atoms Answers

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