

Honors Chemistry - Chapter 12 Test

True/False - Clearly indicate the correct choice

1. In an aqueous solution of table salt (NaCl), the table salt would be considered the solvent.
False
2. Water and gasoline are two liquids that would be considered miscible.
False
3. A negative sign listed for the enthalpy of solution means that energy is released in the formation of the solution.
True
4. A chemist finds that increasing the temperature leads to less of a substance dissolving in water. The substance is most likely a gas.
True
5. Salt or sugar dissolved in water is considered to be a homogeneous mixture.
True

Matching Set - Clearly indicate the correct choice

- a. solution b. colloid c. suspension
6. largest solute particle size c
 7. cannot be separated by filtration, will not scatter light a
 8. heterogeneous, does not separate on standing c
 9. particles settle out a
 10. smallest solute particle size b
 11. muddy water c
 12. shaving cream b

Multiple Choice - Clearly indicate the correct choice

13. A solute dissolved in alcohol is known as a/an
a. aqueous solution b. heterogeneous mixture c. tincture
d. immiscible solution
c
14. An electrolyte is a substance that dissolves in water to form a solution

Modern Chemistry Chapter 12 Test

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Modern Chemistry Chapter 12 Test:

Unveiling the Magic of Words: A Review of "**Modern Chemistry Chapter 12 Test**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Modern Chemistry Chapter 12 Test**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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