
AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5. $q=mc(\Delta T)$; What does "q" stand for?: Heat (kJ)
6. $q=mc(\Delta T)$; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12. $\text{pH} < 7$; acid, base, or neutral?: Acid
13. $\text{pH} > 7$; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt, $\text{pH} = 7$.
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

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JL Elias



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Whispering the Strategies of Language: An Emotional Quest through **Pre Ap Chemistry Spring Final Review Tenpayore**

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Table of Contents Pre Ap Chemistry Spring Final Review Tenpayore

1. Understanding the eBook Pre Ap Chemistry Spring Final Review Tenpayore
 - The Rise of Digital Reading Pre Ap Chemistry Spring Final Review Tenpayore
 - Advantages of eBooks Over Traditional Books
2. Identifying Pre Ap Chemistry Spring Final Review Tenpayore
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pre Ap Chemistry Spring Final Review Tenpayore
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pre Ap Chemistry Spring Final Review Tenpayore

- Personalized Recommendations
 - Pre Ap Chemistry Spring Final Review Tenpayore User Reviews and Ratings
 - Pre Ap Chemistry Spring Final Review Tenpayore and Bestseller Lists
5. Accessing Pre Ap Chemistry Spring Final Review Tenpayore Free and Paid eBooks
 - Pre Ap Chemistry Spring Final Review Tenpayore Public Domain eBooks
 - Pre Ap Chemistry Spring Final Review Tenpayore eBook Subscription Services
 - Pre Ap Chemistry Spring Final Review Tenpayore Budget-Friendly Options
 6. Navigating Pre Ap Chemistry Spring Final Review Tenpayore eBook Formats
 - ePub, PDF, MOBI, and More
 - Pre Ap Chemistry Spring Final Review Tenpayore Compatibility with Devices
 - Pre Ap Chemistry Spring Final Review Tenpayore Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pre Ap Chemistry Spring Final Review Tenpayore
 - Highlighting and Note-Taking Pre Ap Chemistry Spring Final Review Tenpayore
 - Interactive Elements Pre Ap Chemistry Spring Final Review Tenpayore
 8. Staying Engaged with Pre Ap Chemistry Spring Final Review Tenpayore
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pre Ap Chemistry Spring Final Review Tenpayore
 9. Balancing eBooks and Physical Books Pre Ap Chemistry Spring Final Review Tenpayore
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pre Ap Chemistry Spring Final Review Tenpayore
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Pre Ap Chemistry Spring Final Review Tenpayore
 - Setting Reading Goals Pre Ap Chemistry Spring Final Review Tenpayore
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Pre Ap Chemistry Spring Final Review Tenpayore

- Fact-Checking eBook Content of Pre Ap Chemistry Spring Final Review Tenpayore
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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