

Chapter 4 – Atomic Structure

Short Answer: Answer the following questions in complete sentences using science vocabulary. Make sure that your answers are complete and detailed.

1. A sample of calcium contains calcium-40, calcium-44, calcium-42, calcium-48, calcium-43, and calcium-46 atoms. Explain why these atoms have different mass numbers but must have the same atomic number.

The atoms have different mass numbers because they have a different number of neutrons. The mass number is equaled to the number of protons plus neutrons.

The atoms must have the same atomic number because the atomic number is equaled to the number of protons. The number of protons is unique for each element.

2. Complete the following table. Circle the name of the element(s) that are isotopes. Complete Bohr's Model diagrams for the element with eleven electrons and the element with an atomic number of 9.

| Element Name | Element Symbol | Atomic Number | Mass Number | Protons | Neutrons | Electrons |
|---------------|----------------|---------------|-------------|---------|----------|-----------|
| Scandium | Sc | 21 | 45 | 21 | 24 | 21 |
| Fluorine | F | 9 | 19 | 9 | 10 | 9 |
| Sodium | Na | 11 | 24 | 11 | 13 | 11 |
| Answers Vary | | | | | 82 | |

Chapter 4 Atomic Structure Answers

Singhal Atul



Chapter 4 Atomic Structure Answers :

Decoding **Chapter 4 Atomic Structure Answers** : Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 4 Atomic Structure Answers** ," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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