

## Chapter 6 Covalent Bonds

**Section 6.2 Covalent Bonding****(pages 166–188)**

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

**Reading Strategy (page 166)**

**Relating Text and Visuals** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron-dot diagrams	Dots represent valence electrons.
Structural formula	
Space-filling	
Electron cloud	

**Covalent Bonds (pages 166–187)**

- Define a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- A molecule is a \_\_\_\_\_ group of atoms that are joined together by one or more covalent bonds. Circle the correct answer.  
negative      neutral      positive
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - one
  - two
  - three

# Chapter 6 Chemical Bonding Section 2 Covalent Answer Key

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**Chapter 6 Chemical Bonding Section 2 Covalent Answer Key:**

## Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Chapter 6 Chemical Bonding Section 2 Covalent Answer Key**

In some sort of inundated with screens and the cacophony of immediate transmission, the profound energy and mental resonance of verbal beauty often disappear in to obscurity, eclipsed by the constant barrage of sound and distractions. Yet, situated within the musical pages of **Chapter 6 Chemical Bonding Section 2 Covalent Answer Key**, a interesting perform of literary elegance that pulses with organic emotions, lies an unforgettable trip waiting to be embarked upon. Composed by a virtuoso wordsmith, this exciting opus manuals readers on an emotional odyssey, softly exposing the latent possible and profound impact stuck within the complicated internet of language. Within the heart-wrenching expanse of the evocative analysis, we shall embark upon an introspective exploration of the book is central themes, dissect their interesting publishing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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