

Chemical Equations and Reactions

Section 2

Concepts & Examples

Balance the following equations for the space provided.

1. Match the product on the left with an appropriate description on the right.

- | | |
|----------------------------|---------------------------------|
| _____ CaCO_3 | (a) a precipitate forms. |
| _____ H_2 | (b) a gas forms. |
| _____ H_2O | (c) a red precipitate forms. |
| _____ CaCl_2 | (d) a white precipitate forms. |
| _____ Ca(OH)_2 | (e) a chemical reaction occurs. |
| _____ Ca | (f) a chemical reaction occurs. |

2. From the following list, select the following equation:



3. In each of the following reactions, write the correct number of products.

- _____ $\text{Ca} + \text{H}_2\text{O} \rightarrow$
- _____ $\text{H}_2 + \text{O}_2 \rightarrow$
- _____ $\text{Ca} + \text{HCl} \rightarrow$
- _____ $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$

4. Balance each following equation.



Chapter 8 Review Chemical Equations Answer

Jicheng Xie



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