

## CHEM 101 Final Exam with complete solutions

How many significant figures are in this number, 0.01160 - ANSWER: → 4

Select the correct scientific notation for 117000 - ANSWER: →  $1.17 \times 10^5$

the temperature of an experiment is designed at 122 degrees F. What is it in Celsius? - ANSWER: → 50 degrees Celsius

An experiment is performed at -173 degrees C. What is the temperature in Kelvin? - ANSWER: → 100K

A piece of metal has a mass of 340 g and a volume of 17.6 mL. What is the density of the metal? - ANSWER: → 19.3 g/mL

The comparison of the density of a substance with the density of water is defined as \_\_\_\_\_. - ANSWER: → specific gravity

The following forms of energy can be classified as potential energy EXCEPT \_\_\_\_\_.

A). Gasoline B). Running

C). Peanut butter sandwich D). Water held by a dam - ANSWER: → running

The following examples are chemical changes EXCEPT

A). Rusting iron B). Cooking an egg

C). Burning gasoline D). Melting ice - ANSWER: → melting ice

The amount of heat necessary to raise the temperature of 1 g liquid water by 1°C, is

defined as \_\_\_\_\_. - ANSWER: → a calorie

# Chem 101 Final Exam Answers

**S Marginson**



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