

ENHANCING TEST ITEMS – CHEMISTRY (for Physical Sciences majors)

1. Which of the following statements best describes the atomic radius of potassium atom and the atomic radius of a calcium atom?
- The radius of the potassium atom is smaller because of its smaller nuclear charge.
 - The radius of the potassium atom is smaller because of its larger nuclear size.
 - The radius of the potassium atom is larger because its nuclear charge is smaller.
 - The radius of the potassium atom is larger because its nuclear charge is also larger.

2. Which type of chemical reaction occurs when an iron nail rusts?
- condensation
 - neutralization
 - oxidation-reduction
 - ionization-dissociation

3. Which electron dot symbol represents the atom in period-4 with the highest first ionization energy?

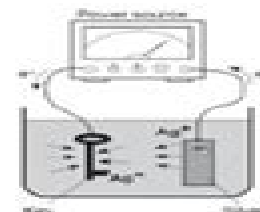


4. Which phase change is exothermic?



5. Given the diagram, which statement BEST describes the key?

- It is negative and acts as the cathode.
- It is positive and acts as the cathode.
- It is negative and acts as the anode.
- It is positive and act as the anode



6. Which technique is safest for diluting a concentrated acid with water?

- Add the acid to the water quickly.
- Add the water to the acid quickly.
- Add the acid to the water slowly while stirring constantly.
- Add the water to the acid slowly while stirring constantly.

7. Which elements combine to form an ionic bond?

- sodium and potassium
- sodium and oxygen
- carbon and oxygen
- carbon and sulfur

8. As the elements in group 16 are considered in order from top to bottom, the atomic radius of each successive element increases. This increase is primarily due to an increase in

- electronegativity
- ionization energy
- the number of protons occupying the nucleus
- the number of occupied principal energy levels

9. An air pollution situation in which reactions occur on the surfaces of solid particles and produce acidic irritation of lung tissues is called _____.

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Yijin Wang



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