

# Solution Stoichiometry



$$\frac{0.29 \text{ mol KI}}{\cancel{L}} \times \frac{0.037 \cancel{L}}{1} \times \frac{1 \text{ mol Pb}(\text{NO}_3)_2}{2 \text{ mol KI}} \times \frac{1}{0.042 \text{ L}}$$

$$= 0.1277 \text{ M}$$

$$\text{M} = \frac{\text{mol}}{\text{L}}$$

# Chemistry Solution Stoichiometry

**LM Reder**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

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