

### balance nuclear reaction equations

aluminum-26 undergoes positron emission  
 potassium-40 undergoes beta decay (electron)  
 uranium-238 undergoes alpha decay



### electron configurations

Write the electron configuration for C.  
 Write the electron configuration for Cr.  
 Write the shorthand notation for Bi.



### NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride  $\text{SF}_6$
- b. Lithium nitride  $\text{Li}_3\text{N}$
- c. Chromium (III) carbonate  $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride  $\text{SnCl}_2$
- e. Ammonium acetate  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  (or  $\text{CH}_3\text{COONH}_4$ )
- f. Mercury (I) chloride  $\text{Hg}_2\text{Cl}_2$
- g. Potassium bromate  $\text{KBrO}_3$
- h. Hydrosulfuric acid  $\text{H}_2\text{S(aq)}$
- i. Chloric acid  $\text{HClO}_3$
- j. Sulfurous acid  $\text{H}_2\text{SO}_3$

Name each of the following compounds:

- a.  $\text{CuSO}_4$  Copper (II) sulfate
- b.  $\text{AlF}_3$  aluminum fluoride
- c.  $\text{HI}$  hydrogen iodide
- d.  $\text{NO}$  nitrogen monoxide
- e.  $\text{H}_2\text{Se(aq)}$  hydroselenic acid
- f.  $\text{HNO}_3$  nitric acid
- g.  $\text{HNO}_2$  nitrous acid
- h.  $\text{NaHSO}_4$  sodium hydrogen sulfate

### Half-life

Tritium ( $\text{H-3}$ ) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?



Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?



### Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.



# Modern Chemistry Chapter 6 Review Answers

**Minjie Lin**



## **Modern Chemistry Chapter 6 Review Answers:**

*Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001      *Foundations of College Chemistry, International Adaptation* Morris Hein, Susan Arena, Cary Willard, 2023-05-16

*Foundations of College Chemistry* 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students. It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence. The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need. This International Adaptation offers new and updated content with improved presentation of all course material. It builds on the strengths of previous editions including clear explanations and step by step problem solving. The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry. There is new and expanded coverage of polarizing power and polarizability, Fajans rules, collision number and mean free path, abnormal molecular masses and van t Hoff factor and applications of radioactivity.

*Organic Synthesis* Michael Smith, 2011-07-12

A reactions oriented course is a staple of most graduate organic programs and synthesis is taught either as a part of that course or as a special topic. Ideally the incoming student is an organic major who has a good working knowledge of basic reactions, stereochemistry and conformational principles. In fact however many often most of the students in a first year graduate level organic course have deficiencies in their undergraduate work, are not organic majors and are not synthetically inclined. To save students much time catching up this text provides a reliable and readily available source for background material that will enable all graduate students to reach the same high level of proficiency in organic chemistry. Produced over many years with extensive feedback from students taking an organic chemistry course this book provides a reaction based approach. The first two chapters provide an introduction to functional groups, these are followed by chapters reviewing basic organic transformations e.g. oxidation, reduction. The book then looks at carbon-carbon bond formation reactions and ways to disconnect a bigger molecule into simpler building blocks. Most chapters include an extensive list of questions to test the reader's understanding. There is also a new chapter outlining full retrosynthetic analyses of complex molecules which highlights common problems made by scientists. The book is intended for graduate and postgraduate students, scientific researchers in chemistry. New publisher, new edition, extensively updated and corrected. Over 950 new references with more than 6100 references in total. Over 600 new reactions and figures replaced or updated. Over 300 new homework problems from the current literature to provide nearly 800 problems to test reader understanding of the key principles.

**Medicinal Chemistry** Norma K Dunlap, Donna M Huryn, 2025-11-28

This book provides a concise introduction to the medicinal chemistry of therapeutically active compounds. Written in an accessible style it offers a brief overview of the drug development process from discovery to approved drug. Emphasizing medicinal chemistry strategies the content is organized

by target as well as therapeutic area drug classes It covers existing drugs and the discovery process that led to their development Aimed at students and professionals interested in drug discovery Medicinal Chemistry requires little prior knowledge making it ideal for classroom and training seminar use Written by experts in the field this outstanding new edition explains medicinal chemistry on chemical mechanism terms allowing for a more interesting and user friendly text Includes valuable insights toward the various pathways taken by the pharmaceutical industry in drug discoveries Enhanced text includes questions raised and suggestions made from students in the authors medicinal chemistry class Content provided to enhance the text includes a solutions manual PowerPoint slides for lecture outlines and especially for the earlier chapters a selection of videos An accompanying website will be updated to include additional material 1 Journal Club exercises current journal articles with guiding questions for student analysis two per chapter 2 additional case studies 3 organic synthesis analysis selected syntheses both discovery and process will be discussed and 4 updates on current advances in the field of medicinal chemistry The book benefits a diverse audience including upper level undergraduates and graduates who are studying in the fields of medicinal chemistry and drug discovery as well as scientists working in the pharmaceutical industry

*Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Study Guide to accompany Basic Concepts of Chemistry, 7th Edition* Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry

**Molecular Chemistry and Biomolecular Engineering** Lionello Pogliani, Francisco Torrens, A. K. Haghi, 2019-09-12 This new volume is devoted to molecular chemistry and its applications to the fields of biology It looks at the integration of molecular chemistry with biomolecular engineering with the goal of creating new biological or physical properties to address scientific or societal challenges It takes a both multidisciplinary and interdisciplinary perspective on the interface between molecular biology biophysical chemistry and chemical engineering *Molecular Chemistry and Biomolecular Engineering Integrating Theory and Research with Practice* provides effective support for the development of the laboratory and data analysis skills that researchers will draw on time and again for the practical aspects and also gives a solid grounding in the broader transferable skills *Basic Concepts of Chemistry, Study Guide* Leo J. Malone, 2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome

the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems **Math Review Toolkit for General Chemistry** Gary Long,2006-05-30

Modern University Chemistry, Custom Pub Norbert T. Porile,1993 **Technical Book Review Index** ,1965

**Introduction to Modern Chemistry: Student Manual** Eugene Meyer,1979 **Athenaeum and Literary Chronicle** ,1871 Books in Print Supplement ,2002 **The Athenaeum** ,1871 **Water and Gas Review** ,1915 *India Rubber World and Electrical Trades Review* John Robertson Dunlap, Henry Clemens Pearson,1921 Children's Books in Print R R Bowker Publishing, Bowker,1999-12 The Australian Mathematics Teacher ,1998 **The Athenaeum** James Silk Buckingham, John Sterling, Frederick Denison Maurice, Henry Stebbing, Charles Wentworth Dilke, Thomas Kibble Hervey, William Hepworth Dixon, Norman Maccoll, Vernon Horace Rendall, John Middleton Murry,1871

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