

# MOLE TO MOLE STOICHIOMETRY WORKSHEET

Name : \_\_\_\_\_

Date : \_\_\_\_\_

1. Using the equation



- Find the number of moles of Al used to produce 9.9 moles of  $\text{H}_2$ .
- Find the number of moles of HCl used to produce 1.11 moles of  $\text{AlCl}_3$ .
- Find the number of moles of  $\text{AlCl}_3$  produced along with 0.0757 moles of  $\text{H}_2$ .

2. The reaction between lithium and fluorine produces lithium fluoride. How many moles of lithium fluoride can be produced with 2.17 mol of fluorine?



3. In the reaction of copper (II) oxide breaking down to form copper and oxygen, how much copper is produced if 0.663 moles of oxygen is given off?



4. Fermentation is a complex chemical process of making wine by converting glucose into ethanol and carbon dioxide:



If 17.5 moles of ethanol were produced, how many moles of glucose were there initially?

5. Consider the reaction:



Calculate the moles of HCl needed to react completely with 8.25 moles of zinc.

# Moles And Stoichiometry Practice Problems Answers

**Sabine Zange**



## **Moles And Stoichiometry Practice Problems Answers:**

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

**A Visual Analogy Guide to Chemistry, 2e** Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

**Basics for Chemistry** David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

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**Improving Student Comprehension in Chemistry Laboratories** Tracy Lynn Haroff, 2006

**Holt Chemistry** Ralph Thomas Myers, 2004

**Practice Makes Perfect Chemistry Review and Workbook, Second Edition** Marian DeWane, Heather Hattori, 2018-12-28

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*Prentice Hall Chemistry*, 2000

**Chemistry** John McMurry, 1998

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*Exploring Chemistry (Loose-Leaf)* Matthew Johll,2012-05-04 Matthew Johll's Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It's the same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book's specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

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