

Acids, Bases, Buffers and pH Data Tables

Part A.

	Solution	pH Meter		pH paper		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value	color	pH value
	Control (H_2O)						
a.	0.1 M NH_3	11.00	$10^{-11} = 1^{-11}$	blue	12	purple	10
b.	0.1 M NaOH	12.85	$10^{-12.85} = 1.41^{-13}$	blue	13	purple	10
c.	0.1 M H_3BO_3	6.5	$10^{-6.5} = 3.16^{-7}$	orange yellow	4.5	yellow	6.5
d.	0.1 M HCl	1.40	$10^{-1.40} = .039$	orange	3	red/pink	3

Part B.

	Solution	pH Meter		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value
	Control (H_2O)				
a.	0.1 M Na_2CO_3	11.42	$10^{-11.42} = 3.80^{-12}$	purple	10
b.	0.1 M $(NH_4)_2SO_4$	6.58	$10^{-6.58} = 2.63^{-7}$	yellow	6
c.	0.1 M NaCl	6.80	$10^{-6.80} = 1.58^{-7}$	bright green	6.5

Part C.

	Solution	pH Meter
		Reading
	Control (H_2O)	
a.	Acetic Acid	3.05
b.	Sodium Acetate	7.15
c.	Mixture of a & b	4.83

Ph Properties Of Buffer Solutions Pre Lab Answers

S Marginson



Ph Properties Of Buffer Solutions Pre Lab Answers:

Reviewing **Ph Properties Of Buffer Solutions Pre Lab Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Ph Properties Of Buffer Solutions Pre Lab Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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