

Molecular Composition of Gases

PROBLEM 10

PROBLEM 10. Answer the following questions in the space provided.

1. The composition of a gas mixture is most easily given in the _____
 - (a) volume of each component
 - (b) pressure of the gas
 - (c) composition of the gas
 - (d) number of molecules per unit volume
2. Molar fractions are (a) _____, (b) _____, (c) _____, (d) _____, (e) _____, (f) _____, (g) _____, (h) _____, (i) _____, (j) _____, (k) _____, (l) _____, (m) _____, (n) _____, (o) _____, (p) _____, (q) _____, (r) _____, (s) _____, (t) _____, (u) _____, (v) _____, (w) _____, (x) _____, (y) _____, (z) _____.
3. The composition of a gas mixture is most easily given in the _____
 - (a) volume of each component
 - (b) pressure of the gas
 - (c) composition of the gas
 - (d) number of molecules per unit volume
4. The composition of a gas mixture is most easily given in the _____
 - (a) volume of each component
 - (b) pressure of the gas
 - (c) composition of the gas
 - (d) number of molecules per unit volume
5. The composition of a gas mixture is most easily given in the _____
 - (a) volume of each component
 - (b) pressure of the gas
 - (c) composition of the gas
 - (d) number of molecules per unit volume

PROBLEM 11. Answer the following questions in the space provided.

1. The composition of a gas mixture is most easily given in the _____

Chapter 11 Review Molecular Composition Of Gases

Mixed Answers

David Kirk



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