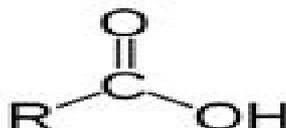


Please answer each question carefully. Remember, follow your gut! Please copy all answers on the answer sheet on the back of this exam! Otherwise, I will take off 10 points!

- The structure of DNA described by Watson and Crick included:
  - a double helix
  - the sugar-phosphate backbone aligned in the center of the helix
  - the base pairs that are stacked on the inside of the double helix
  - A double helix and base pairs that are stacked on the inside of the double helix
- What gives proteins such a dominant role in biochemistry?
  - the ability to act as a blueprint
  - their ability to spontaneously fold into complex three-dimensional structures
  - the variation in protein sizes
  - their ability to self-replicate
- Match the loss of a particular organelle with the associated disease.
  - Hypercholesterolemia - smooth endoplasmic reticulum
  - Diabetes - endosome
  - Tay-Sachs disease - lysosome
  - Muscle degeneration - mitochondria
- In a biochemistry lab course, you are asked to design an experiment to identify a strain of bacteria. Your lab partner claims that she thinks the bacterium contains a rough endoplasmic reticulum. To verify her claim, which of the following experiments would you perform? (pretend you're indulging her ludicrous claim).
  - determine whether the bacterium has an internal membrane-enclosed compartment
  - determine whether the bacterium generates  $\text{CO}_2$  in the presence of fuel molecules
  - determine whether the bacterium can synthesize proteins
  - determine whether the bacterium can synthesize ATP in the presence of fuel molecules and  $\text{O}_2$
- What is the  $\text{H}^+$  concentration in a urine sample that has a pH of 6?
  - $10^{-6} \text{ M}$
  - $10^{-8} \text{ M}$
  - $10^6 \text{ M}$
  - $10^{-14} \text{ M}$
- Which of the following is considered a noncovalent bond?
  - electrostatic interactions
  - hydrogen bonds
  - van der Waals interactions
  - All of these answers
- Which functional group is shown below?



- Amine
  - Thiol
  - Carboxylic acid
  - None of these answers
- What is the  $[\text{A}^-]/[\text{HA}]$  ratio when the weak acid is in a solution one pH unit above its  $\text{pK}_a$ ?

# Biochemistry Quiz Questions And Answers

**Angus Stevenson**



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*Human Physiology, Biochemistry and Basic Medicine* Laurence A. Cole, Peter R. Kramer, 2015-10-13 Human Physiology Biochemistry and Basic Medicine is a unique perspective that draws together human biology physiology biochemistry nutrition and cell biology in one comprehensive volume In this way it is uniquely qualified to address the needs of the emerging field of humanology a holistic approach to understanding the biology of humans and how they are distinguished from other animals Coverage starts with human anatomy and physiology and the details of the workings of all parts of the male and female body Next coverage of human biochemistry and how sugars fats and amino acids are made and digested is discussed as is human basic medicine covering the science of diseases and human evolution and pseudo evolution The book concludes with coverage of basic human nutrition diseases and treatments and contains broad coverage that will give the reader an understanding of the entire human picture Covers the physiology anatomy nutrition biochemistry and cell biology of humans showing how they are distinguished from other animals Includes medical literature and internet references example test questions and a list of pertinent words at the end of each chapter Provides unique perspective into all aspects of what makes up and controls humans

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