

## Chapter Test A

### Chapter: Chemical Equations and Reactions

Do the work provided below the letter of the item or place the item completion mark underneath or beside answers with questions.

1. Which statement is **not** correct?
  - a. The number of atoms is conserved.
  - b. The number of molecules is not conserved.
  - c. The number of molecules is not conserved, but the number of atoms is conserved.
  - d. None of the above; all three are correct.
2. In the reaction described by a chemical equation,
 
$$\text{reactants} \rightarrow \text{products} + \text{products} + \text{products} + \text{products}$$
  - a. all the reactants are products.
  - b. all the reactants are products.
  - c. all the reactants are products.
  - d. None is correct.
3. Which of the following relationships is **not** correct?
  - a. The products of a reaction are the substances with mass (g) listed below of the equation.
  - b. All of the reactants are products.
  - c. The reactants of a reaction are the substances listed above the equation.
  - d. The reactants of a reaction are the substances listed below the equation.
4. Which is **not** a correct statement?
  - a. The number of atoms is conserved.
  - b. The number of molecules is not conserved.
  - c. The number of atoms is not conserved.
  - d. All of the above.
5. In a reaction, a chemical equation tells us
  - a. about the substances.
  - b. about the number of products.
  - c. about the number of reactants.
  - d. about the coefficients.

# Chapter 9 Modern Chemistry Test

**John J. Lentini**



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