

CHEM 101 Final Exam with complete solutions

How many significant figures are in this number, 0.01160 - ANSWER: → 4

Select the correct scientific notation for 117000 - ANSWER: → 1.17×10^5

the temperature of an experiment is designed at 122 degrees F. What is it in Celsius? - ANSWER: → 50 degrees Celsius

An experiment is performed at -173 degrees C. What is the temperature in Kelvin? - ANSWER: → 100K

A piece of metal has a mass of 340 g and a volume of 17.6 mL. What is the density of the metal? - ANSWER: → 19.3 g/mL

The comparison of the density of a substance with the density of water is defined as _____. - ANSWER: → specific gravity

The following forms of energy can be classified as potential energy EXCEPT _____.

- A). Gasoline B). Running
- C). Peanut butter sandwich D). Water held by a dam - ANSWER: → running

The following examples are chemical changes EXCEPT

- A). Rusting iron B). Cooking an egg
- C). Burning gasoline D). Melting ice - ANSWER: → melting ice

The amount of heat necessary to raise the temperature of 1 g liquid water by 1°C, is

defined as _____. - ANSWER: → a calorie

Chem 101 Final Exam Answers

Camilla Rothe



Chem 101 Final Exam Answers:

Chem 101 Final Exam Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chem 101 Final Exam Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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