

QUESTION 1

Volume of pipette used 25.00 cm^3
 (V₀)

Titration	Rough	1	2	3
Final burette reading (cm ³)	20.80	41.40	20.40	40.90
Initial burette reading (cm ³)	0.00	20.80	0.00	20.40
Volume of A used	20.80	20.60	20.40	20.50

② ^{average} Volume of A used = $\frac{20.60 + 20.40 + 20.50}{3}$
 $= 20.50 \text{ cm}^3$

③(i) Concentration of A in mol/dm³ = $\frac{\text{conc. in g/dm}^3}{\text{molar mass}}$
 $= \frac{4.4 \text{ g/dm}^3}{2[13] + 32 + 4[16]}$
 $= \frac{4.4}{98} = 0.0449 \text{ mol/dm}^3$

③(ii) Concentration of B in mol/dm³, C_B

Using $\frac{C_A V_A}{C_B V_B} = \frac{n_A}{n_B}$

$$\frac{0.0449 \times 20.50}{C_B \times 25} = \frac{1}{1}$$

$$C_B = \frac{0.0449 \times 20.50}{25} = 0.03287 \text{ mol/dm}^3$$

Chemistry Practical Question And Answer

Robert R. Redfield



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Table of Contents Chemistry Practical Question And Answer

1. Understanding the eBook Chemistry Practical Question And Answer
 - The Rise of Digital Reading Chemistry Practical Question And Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practical Question And Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Practical Question And Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practical Question And Answer
 - Personalized Recommendations
 - Chemistry Practical Question And Answer User Reviews and Ratings
 - Chemistry Practical Question And Answer and Bestseller Lists
5. Accessing Chemistry Practical Question And Answer Free and Paid eBooks
 - Chemistry Practical Question And Answer Public Domain eBooks
 - Chemistry Practical Question And Answer eBook Subscription Services

- Chemistry Practical Question And Answer Budget-Friendly Options
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 - Chemistry Practical Question And Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Practical Question And Answer
 - Highlighting and Note-Taking Chemistry Practical Question And Answer
 - Interactive Elements Chemistry Practical Question And Answer
- 8. Staying Engaged with Chemistry Practical Question And Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Practical Question And Answer
- 9. Balancing eBooks and Physical Books Chemistry Practical Question And Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Practical Question And Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Practical Question And Answer
 - Setting Reading Goals Chemistry Practical Question And Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Practical Question And Answer
 - Fact-Checking eBook Content of Chemistry Practical Question And Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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