

Introduction 11

An Electron's Address

Name: _____
 Date: _____
 Class: _____

Introduction: Energy Levels and Sublevels

In previous lessons you have learned about the Bohr model of the atom. The Bohr model of the atom says that electrons orbit the nucleus in fixed energy levels. The Bohr model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels.

Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels.

Table 1

Energy Level	Number of Sublevels in Each Energy Level
1	1
2	2
3	3
4	4

What does Bohr's model of the atom say about the number of sublevels in each energy level? Bohr's model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.

Critical Thinking Questions

- How many sublevels are in the 1st energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.
- How many sublevels are in the 2nd energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.
- How many sublevels are in the 3rd energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.
- How many sublevels are in the 4th energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.
- How many sublevels are in the 5th energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.
- How many sublevels are in the 6th energy level?
 The Bohr model of the atom says that the number of sublevels in each energy level is equal to the square of the principal quantum number.

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Ensheng Dong



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Database, 1989 **Comprehensive Medicinal Chemistry** Corwin Hansch, Peter George Sammes, John Bodenhan Taylor, 1990 Drug design is a multi disciplinary activity involving chemists biologists biochemists pharmacologists and many others the chemist's role is central in inventing new compounds which exert a beneficial effect However once a lead for a new active drug has been established its effective delivery has to be demonstrated and extensive toxicological studies undertaken to demonstrate its safety before clinical trials can commence The metabolic fate of the drug has to be revealed and detailed distribution studies carried out in order to satisfy the regulatory authorities before the new compound can be marketed Comprehensive Medicinal Chemistry describes all these aspects of the design of a drug whilst centering on the chemical mechanism whereby such agents act Volume 4 covers quantitative drug design Online, 1989 Chemical Week, 2000 **Detroit Suburban North Woodward Area Telephone Directories**, 1994 **The Directory of Management Consultants, 2003** Kennedy Information (Firm), 2002 *Book of Abstracts* American Chemical Society. Meeting, American Chemical Society, 1986

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