

FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

| Elements | metal ion | non-metal ion | Chemical Formula | Name |
|-----------------------|-----------|---------------|------------------|-------------------|
| potassium and iodine | K^{+1} | I^{-1} | KI | Potassium Iodide |
| barium and chlorine | Ba^{2+} | Cl^{-1} | $BaCl_2$ | Barium Chloride |
| sodium and oxygen | Na^{+1} | O^{2-} | Na_2O | sodium Oxide |
| aluminum and nitrogen | Al^{3+} | N^{3-} | AlN | Aluminum Nitride |
| lithium and sulfur | Li^{+1} | S^{2-} | Li_2S | Lithium Sulfide |
| magnesium and oxygen | Mg^{+2} | O^{2-} | MgO | Magnesium Oxide |
| calcium and chlorine | Ca^{2+} | Cl^{-1} | $CaCl_2$ | calcium chloride |
| boron and chlorine | B^{3+} | Cl^{-1} | BCl_3 | boron chloride |
| beryllium and sulfur | Be^{2+} | S^{2-} | BeS | beryllium sulfide |

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

| | F | Cl | O^{2-} | S^{2-} | N^{3-} | P^{3-} |
|-----------|---------|----------|-----------|-----------|-----------|-----------|
| Li^{+} | LiF | $LiCl$ | Li_2O | Li_2S | Li_3N | Li_3P |
| Na^{+} | NaF | $NaCl$ | Na_2O | Na_2S | Na_3N | Na_3P |
| K^{+} | KF | KCl | K_2O | K_2S | K_3N | K_3P |
| Be^{2+} | BeF_2 | $BeCl_2$ | BeO | BeS | Be_3N_2 | Be_3P_2 |
| Mg^{2+} | MgF_2 | $MgCl_2$ | MgO | MgS | Mg_3N_2 | Mg_3P_2 |
| Ca^{2+} | CaF_2 | $CaCl_2$ | CaO | CaS | Ca_3N_2 | Ca_3P_2 |
| Al^{3+} | AlF_3 | $AlCl_3$ | Al_2O_3 | Al_2S_3 | AlN | AlP |
| Ga^{3+} | GaF_3 | $GaCl_3$ | Ga_2O_3 | Ga_2S_3 | GaN | GaP |
| B^{3+} | BF_3 | BCl_3 | B_2O_3 | B_2S_3 | BN | BP |

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Table of Contents Ionic Formula Writing Kit Answers

1. Understanding the eBook Ionic Formula Writing Kit Answers
 - The Rise of Digital Reading Ionic Formula Writing Kit Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Formula Writing Kit Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Formula Writing Kit Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Formula Writing Kit Answers
 - Personalized Recommendations
 - Ionic Formula Writing Kit Answers User Reviews and Ratings
 - Ionic Formula Writing Kit Answers and Bestseller Lists
5. Accessing Ionic Formula Writing Kit Answers Free and Paid eBooks
 - Ionic Formula Writing Kit Answers Public Domain eBooks
 - Ionic Formula Writing Kit Answers eBook Subscription Services
 - Ionic Formula Writing Kit Answers Budget-Friendly Options

6. Navigating Ionic Formula Writing Kit Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Formula Writing Kit Answers Compatibility with Devices
 - Ionic Formula Writing Kit Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Formula Writing Kit Answers
 - Highlighting and Note-Taking Ionic Formula Writing Kit Answers
 - Interactive Elements Ionic Formula Writing Kit Answers
8. Staying Engaged with Ionic Formula Writing Kit Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Formula Writing Kit Answers
9. Balancing eBooks and Physical Books Ionic Formula Writing Kit Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Formula Writing Kit Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Formula Writing Kit Answers
 - Setting Reading Goals Ionic Formula Writing Kit Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Formula Writing Kit Answers
 - Fact-Checking eBook Content of Ionic Formula Writing Kit Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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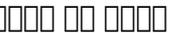
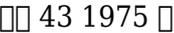
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