

CHEMISTRY
PAPER – 2
(PRACTICAL)

(Maximum Marks: 30)

(Time allowed: Three hours)

*(Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.)*

ALL ANSWERS MUST BE WRITTEN IN THE ANSWER BOOKLET PROVIDED SEPARATELY.

*Question 1 is an oxidation–reduction titration in which sufficient working details are given.
All essential working must be shown.*

*Question 2 is an exercise dealing with identification of organic compounds.
Credit will be given for precise observations recorded and for well–drawn deductions.*

Question 3 is an exercise in qualitative analysis.

Read the questions carefully and follow the given instructions.

Attempt all questions.

All working, including rough work, should be done on the same sheet as the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Mathematical Tables are provided.

Attempt all questions.

Question 1

[7]

You are provided with two solutions as follows:

- C-10 is a solution containing 1.95 gms of potassium manganate (VII), KMnO_4 per litre.
 - C-11 is a solution prepared by dissolving 23.2 gms of hydrated ammonium iron (II) sulphate crystals, $(\text{NH}_4)_2\text{SO}_4 \cdot \text{FeSO}_4 \cdot x\text{H}_2\text{O}$ per litre.
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