

# Chemical Equations and Reactions

## Section 1

### General Knowledge

Answer the following questions for the space provided.

1. Match the formulae on the left with an appropriate description on the right.

- |                         |                                        |
|-------------------------|----------------------------------------|
| _____ $\text{Ca}$       | (A) a gas formed                       |
| _____ $\text{H}_2$      | (B) a precipitate formed               |
| _____ $\text{CO}_2$     | (C) a solid precipitate formed         |
| _____ $\text{CaCO}_3$   | (D) a white solid formed               |
| _____ $\text{CaO}$      | (E) a solid formed in the liquid state |
| _____ $\text{Ca(OH)}_2$ | (F) a solid formed in the liquid state |

2. From the following list, identify the following compound:



3. In the light of the following reactions, write the usual formula of potassium:

- \_\_\_\_\_  $\text{K}_2\text{O}$
- \_\_\_\_\_  $\text{K}_2\text{O}_2$
- \_\_\_\_\_  $\text{K}_2\text{SO}_4$
- \_\_\_\_\_  $\text{K}_2\text{CO}_3$

4. Balance the following equations:



# Chapter 8 Review Chemical Equations Answer

**H Kauffman**



## **Chapter 8 Review Chemical Equations Answer:**

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