



ELECTRONS IN ATOMS

Chapter Test B

A. Matching

Match each item with the correct category in the column that best describes it. Write the letter of the category in the blank.

Column A	Column B
_____ 1. The type of electromagnetic radiation that is most energetic	a. gamma rays
_____ 2. The process of removing electrons from an atom's orbit	b. ionization
_____ 3. The process of moving electrons from one energy level to another	c. photoelectric effect
_____ 4. An experiment that showed that electrons have wave-like properties	d. diffraction
_____ 5. The process of moving electrons into the highest possible energy level	e. photoionization
_____ 6. The process of moving electrons from the lowest possible energy level to the highest possible energy level	f. excitation
_____ 7. The process of moving electrons to lower energy levels from the ground state	g. emission lines
_____ 8. Light that is produced by the transition of electrons from the ground state to an excited state	h. absorption lines
_____ 9. The process of moving electrons from one energy level to another	i. atomic absorption
_____ 10. The process of moving electrons from one energy level to another	j. emission spectrum

B. Multiple Choice

Match each item with the correct answer choice.

_____ 11. The process of moving electrons from one energy level to another	a. ionization	b. photoionization	c. excitation	d. emission lines
_____ 12. The process of moving electrons from one energy level to another	e. absorption lines	f. photoelectric effect	g. diffraction	h. atomic absorption
_____ 13. The process of moving electrons from one energy level to another	i. emission spectrum	j. ionization	k. photoionization	l. excitation
_____ 14. The process of moving electrons from one energy level to another	m. absorption lines	n. photoelectric effect	o. diffraction	p. atomic absorption

Chapter Test B Arrangement Of Electrons In Atoms

Julia Schneider



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