

# Chemistry

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(Chapter 13) (Amines) (Exercises)  
(Class 12)

## Question 13.1:

Write IUPAC names of the following compounds and classify them into primary, secondary and tertiary amines.



## Answer 13.1:

(i) 1-Methylethanamine (1° amine)

(ii) Propan-1-amine (1° amine)

(iii) N-Methyl-2-methylethanamine (2° amine)

(iv) 2-Methylpropan-2-amine (1° amine)

(v) N-Methylbenzamine or N-methylaniline (2° amine)

(vi) N-Ethyl-N-methylethanamine (3° amine)

(vii) 3-Bromobenzenamine or 3-bromoaniline (1° amine)

## Question 13.2:

Give one chemical test to distinguish between the following pairs of compounds.

(i) Methylamine and dimethylamine

(ii) Secondary and tertiary amines

(iii) Ethylamine and aniline

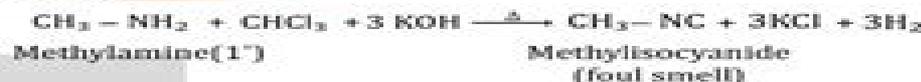
(iv) Aniline and benzylamine

(v) Aniline and N-methylaniline.

## Answer 13.2:

(i) Methylamine and dimethylamine can be distinguished by the carbylamine test.

**Carbylamine test:** Aliphatic and aromatic primary amines on heating with chloroform and ethanolic potassium hydroxide form foul-smelling isocyanides or carbylamines. Methylamine (being an aliphatic primary amine) gives a positive carbylamine test, but dimethylamine does not.



(ii) Secondary and tertiary amines can be distinguished by allowing them to react with Hinsberg's reagent (benzenesulphonyl chloride, C<sub>6</sub>H<sub>5</sub>SO<sub>2</sub>Cl). Secondary amines react with Hinsberg's reagent to form a product that is insoluble in an alkali. For example, N, N-diethylamine reacts with Hinsberg's reagent to form N, N-diethylbenzenesulphonamide, which is insoluble in an alkali. Tertiary amines, however, do not react with Hinsberg's reagent.



(iii) Ethylamine and aniline can be distinguished using the azo-dye test. A dye is obtained when aromatic amines react with HNO<sub>2</sub> (NaNO<sub>2</sub> + dil.HCl) at 0-5°C, followed by a reaction with the alkaline solution of 2-naphthol. The dye is usually yellow, red, or orange in colour. Aliphatic amines give a brisk effervescence due (to the evolution of N<sub>2</sub> gas) under similar conditions.

# Chemistry Chapter 13 Solutions

**Camilla Rothe**



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## **Table of Contents Chemistry Chapter 13 Solutions**

1. Understanding the eBook Chemistry Chapter 13 Solutions
  - The Rise of Digital Reading Chemistry Chapter 13 Solutions
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 13 Solutions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 13 Solutions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 13 Solutions
  - Personalized Recommendations
  - Chemistry Chapter 13 Solutions User Reviews and Ratings

- Chemistry Chapter 13 Solutions and Bestseller Lists
- 5. Accessing Chemistry Chapter 13 Solutions Free and Paid eBooks
  - Chemistry Chapter 13 Solutions Public Domain eBooks
  - Chemistry Chapter 13 Solutions eBook Subscription Services
  - Chemistry Chapter 13 Solutions Budget-Friendly Options
- 6. Navigating Chemistry Chapter 13 Solutions eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 13 Solutions Compatibility with Devices
  - Chemistry Chapter 13 Solutions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 13 Solutions
  - Highlighting and Note-Taking Chemistry Chapter 13 Solutions
  - Interactive Elements Chemistry Chapter 13 Solutions
- 8. Staying Engaged with Chemistry Chapter 13 Solutions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 13 Solutions
- 9. Balancing eBooks and Physical Books Chemistry Chapter 13 Solutions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 13 Solutions
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 13 Solutions
  - Setting Reading Goals Chemistry Chapter 13 Solutions
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 13 Solutions
  - Fact-Checking eBook Content of Chemistry Chapter 13 Solutions
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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