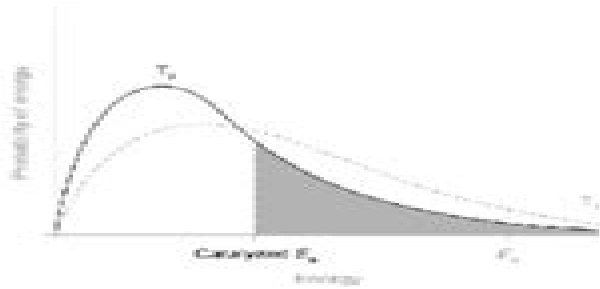


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2.	f	<p>a. Zn^{2+} ions in full shell OR atomic number is same as for neighbouring element</p>	<p>Do not accept Zn in any transition metal. Do not accept alternatives for Zn^{2+} ions.</p>	1
3.	b	<p>ligands donate pairs of electrons to metal ions OR form coordinate covalent bonds</p> <p>ligands are Lewis bases AND metal ions are Lewis acids</p>		1
1.	d	<p>Electrostatic attraction Between oppositely charged ions/between Fe^{2+} and SO_4^{2-}</p>		2
4.	a	 <p>same higher AND to left of T_2 if reactant E_a reduced AND to the left of E_a of curve T_1 if</p>	<p>Do not promote curve into a table, and justify clearly (height and width, or covering a base other than E_a)</p> <p>Do not award AP if curve shown shows significantly more than 10% of molecules with sufficient energy since smaller curve than curve 1.</p> <p>Accept E_a drawn to T_2 instead of curve shown or drawn to left of marked E_a</p>	2
1.	b	<p>to same AND same T_2 if</p>		1

3. Equilibrium favours backward direction/shifts to LHS **AND**

As increase in pressure favours side with less moles of gas.

[1]

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Table of Contents Chemistry Hl Paper 3 M11

1. Understanding the eBook Chemistry Hl Paper 3 M11
 - The Rise of Digital Reading Chemistry Hl Paper 3 M11
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Hl Paper 3 M11
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Hl Paper 3 M11
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Hl Paper 3 M11
 - Personalized Recommendations
 - Chemistry Hl Paper 3 M11 User Reviews and Ratings

- Chemistry HL Paper 3 M11 and Bestseller Lists
- 5. Accessing Chemistry HL Paper 3 M11 Free and Paid eBooks
 - Chemistry HL Paper 3 M11 Public Domain eBooks
 - Chemistry HL Paper 3 M11 eBook Subscription Services
 - Chemistry HL Paper 3 M11 Budget-Friendly Options
- 6. Navigating Chemistry HL Paper 3 M11 eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry HL Paper 3 M11 Compatibility with Devices
 - Chemistry HL Paper 3 M11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry HL Paper 3 M11
 - Highlighting and Note-Taking Chemistry HL Paper 3 M11
 - Interactive Elements Chemistry HL Paper 3 M11
- 8. Staying Engaged with Chemistry HL Paper 3 M11
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry HL Paper 3 M11
- 9. Balancing eBooks and Physical Books Chemistry HL Paper 3 M11
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry HL Paper 3 M11
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry HL Paper 3 M11
 - Setting Reading Goals Chemistry HL Paper 3 M11
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry HL Paper 3 M11
 - Fact-Checking eBook Content of Chemistry HL Paper 3 M11
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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