

CHEMISTRY MIDTERM REVIEW

1.) CONVERSIONS IN METRIC UNITS

How many kilograms are in 143 grams?

Using KHDBDCM

Starting with Grams (g) – base unit. Moving to kilograms (kg); move the decimal three spaces to the right

143 g – 0.143 g

Using conversion factors

KNOW that kilo = 1000 base

$143 \text{ g} \times (1 \text{ kg}/1000\text{g}) = 0.143 \text{ kg}$

2.) CONVERSIONS WITH ENGLISH UNITS

How many milliliters are in 13.5 oz?

MUST USE CONVERSION FACTORS (oz, lb, feet, inches...)

Always start with what you know.... Figure out where to go

13.5 oz

? mL

13.5 oz	1 cup	1 pt	1 qt	1 L	1000 mL
	8 oz	2 cups	2 pt	1.06 qt	1 L

= 398 mL are in 13.5 oz

Question: is ATM (atmospheres) a standard number?

Answer; No, it's a unit. There's not a particular number there. The gas constant has a number and a unit because it is a constant; atmospheres will differ depending on the problem.

Stoichiometry: Practice questions.

22.0 g of propane will combine with what mass (g) of O₂?



Chemistry Midterm Review Answers

James C. Hill



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