

Reference Tables for Physical Setting/CHEMISTRY

2002 Edition

Table A
Standard Temperature and Pressure

Name	Value	Unit
Standard Pressure	101.3 kPa 1 atm	kilopascal atmosphere
Standard Temperature	273 K 0°C	kelvin degree Celsius

Table B
Physical Constants for Water

Heat of Fusion	334 J/g
Heat of Vaporization	2260 J/g
Specific Heat Capacity of H ₂ O (l)	4.18 J/g•°C

Table C
Selected Prefixes

Factor	Prefix	Symbol
10 ³	kilo-	k
10 ⁻¹	deci-	d
10 ⁻²	centi-	c
10 ⁻³	milli-	m
10 ⁻⁶	micro-	μ
10 ⁻⁹	nano-	n
10 ⁻¹²	pico-	p

Table D
Selected Units

Symbol	Name	Quantity
m	meter	length
g	gram	mass
Pa	pascal	pressure
K	kelvin	temperature
mol	mole	amount of substance
J	joule	energy, work, quantity of heat
s	second	time
L	liter	volume
ppm	part per million	concentration
M	molarity	solution concentration

Table E
Selected Polyatomic Ions

H ₃ O ⁺	hydronium	CrO ₄ ²⁻	chromate
Hg ₂ ²⁺	mercurous (II)	Cr ₂ O ₇ ²⁻	dichromate
NH ₄ ⁺	ammonium	MnO ₄ ⁻	permanganate
C ₂ H ₃ O ₂ ⁻ CH ₃ COO ⁻	acetate	NO ₂ ⁻	nitrite
		NO ₃ ⁻	nitrate
CN ⁻	cyanide	O ₂ ²⁻	peroxide
CO ₃ ²⁻	carbonate	OH ⁻	hydroxide
HCO ₃ ⁻	hydrogen carbonate	PO ₄ ³⁻	phosphate
C ₂ O ₄ ²⁻	oxalate	SCN ⁻	thiocyanate
ClO ⁻	hypochlorite	SO ₃ ²⁻	sulfite
ClO ₂ ⁻	chlorite	SO ₄ ²⁻	sulfate
ClO ₃ ⁻	chlorate	H ₂ SO ₄	hydrogen sulfate
ClO ₄ ⁻	perchlorate	S ₂ O ₃ ²⁻	thiosulfate

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Table of Contents Chemistry Reference Table Workbook 2nd Edition Answers

1. Understanding the eBook Chemistry Reference Table Workbook 2nd Edition Answers
 - The Rise of Digital Reading Chemistry Reference Table Workbook 2nd Edition Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Reference Table Workbook 2nd Edition Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Reference Table Workbook 2nd Edition Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Reference Table Workbook 2nd Edition Answers
 - Personalized Recommendations
 - Chemistry Reference Table Workbook 2nd Edition Answers User Reviews and Ratings
 - Chemistry Reference Table Workbook 2nd Edition Answers and Bestseller Lists
5. Accessing Chemistry Reference Table Workbook 2nd Edition Answers Free and Paid eBooks
 - Chemistry Reference Table Workbook 2nd Edition Answers Public Domain eBooks
 - Chemistry Reference Table Workbook 2nd Edition Answers eBook Subscription Services

- Chemistry Reference Table Workbook 2nd Edition Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Chemistry Reference Table Workbook 2nd Edition Answers Compatibility with Devices
 - Chemistry Reference Table Workbook 2nd Edition Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Reference Table Workbook 2nd Edition Answers
 - Highlighting and Note-Taking Chemistry Reference Table Workbook 2nd Edition Answers
 - Interactive Elements Chemistry Reference Table Workbook 2nd Edition Answers
- 8. Staying Engaged with Chemistry Reference Table Workbook 2nd Edition Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Reference Table Workbook 2nd Edition Answers
- 9. Balancing eBooks and Physical Books Chemistry Reference Table Workbook 2nd Edition Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Reference Table Workbook 2nd Edition Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Reference Table Workbook 2nd Edition Answers
 - Setting Reading Goals Chemistry Reference Table Workbook 2nd Edition Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Reference Table Workbook 2nd Edition Answers
 - Fact-Checking eBook Content of Chemistry Reference Table Workbook 2nd Edition Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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